

EMPLOYING DIFFUSION MONTE CARLO IN THE CALCULATION OF MINIMIZED ENERGY PATHS OF THE $\text{CH}_3^+ + \text{H}_2 \leftrightarrow \text{CH}_5^+ \leftrightarrow \text{CH}_3^+ + \text{H}_2$ REACTION AND ITS ISOTOPIC VARIANTS

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Protonated methane is presumed by astrochemists to be an important intermediate in the reaction $\text{CH}_3^+ + \text{HD} \rightarrow \text{CDH}_4^+ \rightarrow \text{CH}_2\text{D}^+ + \text{H}_2$ within the interstellar medium. Understanding this reaction can also help shed light on the observed nonstatistical H/D isotopic abundance in the isotopologues of CH_3^+ within the interstellar medium. Interestingly, based on kinetic studies, Gerlich and co-workers showed that all of the reactions in the series $\text{CH}_{3-n}\text{D}_n^+ + \text{HD} \rightarrow \text{CH}_{4-n}\text{D}_{n+1}^+ \rightarrow \text{CH}_{2-n}\text{D}_{n+1}^+ + \text{H}_2$ have identical net rate constants.^a This result is independent of the value of n.

In previous studies of CH_5^+ , we have employed Diffusion Monte Carlo (DMC)^b to study ground,^c and excited states.^{d,e,f} By performing the simulation in Jacobi coordinates, we can use Adiabatic DMC^g to study the properties of the minimized energy paths of CH_5^+ and isotopologues. To determine the minimized energy path, we calculate the quantum zero-point energy and ground state wave function as a function of the distance between the center of mass of the H_2 group and the center of mass of the CH_3^+ group over a range from 0 to 6 Å. Over this range, we find 5 distinct regions of interaction, short range repulsion region, CH_5^+ complexation, short-range fragment interaction, long-range fragment interaction, and a region of no interaction between the two fragments. Interestingly, the range of H_2/CH_3^+ distances spanned by each of the regions is roughly independent of the number or location of the deuterium atoms. Interestingly, the range of H_2/CH_3^+ distances spanned by each of the regions is roughly independent of the number or location of the deuterium atoms.

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