

## MW SPECTROSCOPY COUPLED WITH ULTRAFAST UV LASER VAPORIZATION: RIBOSE FOUND IN THE GAS PHASE

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Sugars are aldoses or ketoses with multiple hydroxy groups which have been elusive to spectroscopic studies. Here we report a rotational study of the aldopentose ribose. According to any standard textbook aldopentoses can exhibit either linear forms, cyclic five-membered (furanose) structures or six-membered (pyranose) rings, occurring either as  $\alpha$ - or  $\beta$ - anomers depending on the orientation of the hydroxy group at C-1 (anomeric carbon).  $\beta$ -Furanose is predominant in ribonucleosides, RNA, ATP and other biochemically relevant derivatives, but is  $\beta$ -furanose the native form also of free ribose? Recent condensed-phase X-ray<sup>a</sup> and older NMR<sup>b</sup> studies delivered conflicting results. In order to solve this question we conducted a microwave study on D-ribose that, owing to ultrafast UV laser vaporization<sup>c</sup>, has become the first C-5 sugar observed with rotational resolution. The spectrum revealed six conformations of free ribose, preferentially adopting  $\beta$ -pyranose chairs as well as higher-energy  $\alpha$ -pyranose forms. The method also allowed for unambiguous distinction between different orientations of the hydroxy groups, which stabilize the structures by cooperative hydrogen-bond networks. No evidence was observed of the  $\alpha$ -/ $\beta$ -furanoses or linear forms found in the biochemical derivatives.

<sup>a</sup> i) D. Šišak, L. B. McCusker, G. Zandomeneghi, B. H. Meier, D. Bläser, R. Boese, W. B. Schweizer, R. Gylmour and J. D. Dunitz *Angew. Chem. Int. Ed.* **49**, 4503, 2010. ii) W. Saenger *Angew. Chem. Int. Ed.* **49**, 6487, 2010.

<sup>b</sup> i) M. Rudrum, and D. F. Shaw, *J. Chem. Soc.* **52**, 1965. ii) R. U. Lemieux and J. D. Stevens *Can. J. Chem.* **44**, 249, 1966. iii) E. Breitmaier and U. Hollstein *Org. Magn. Reson.* **8**, 573, 1976.

<sup>c</sup> E. J. Cocinero, A. Lesarri, P. Écija, F. J. Basterretxea, J. U. Grabow, J. A. Fernández and F. Castaño *Angew. Chem. Int. Ed.* in press: DOI: 10.1002/anie.201107973, 2012.