

An ideal gas consists of a large number N of identical ultrarelativistic electrons confined to a thin planar region of area A . Use the grand canonical ensemble with chemical potential $\mu = \varepsilon_F$ and temperature 0 to derive its ground state energy U .

(a) Express the sum over orbitals in terms of integrals over position vectors $\mathbf{r} = (x, y)$ and momentum vectors $\mathbf{p} = (p_x, p_y)$.

Express the sum over orbitals in terms of an integral over the orbital energy $\varepsilon = pc$.

$$\begin{aligned} \sum_i 1 &= 2 \frac{\int d^2r \int d^2p}{h^2} = \frac{2A}{h^2} \int 2\pi p dp \\ &= \frac{4\pi A}{h^2} \int \frac{\varepsilon}{c} \frac{d\varepsilon}{c} = \frac{4\pi A}{h^2 c^2} \int \varepsilon d\varepsilon \end{aligned}$$

(b) Express N and U as definite integrals (with upper and lower limits) over ε .

$$\begin{aligned} N &= \frac{4\pi A}{h^2 c^2} \int_0^{\varepsilon_F} \varepsilon d\varepsilon \\ U &= \frac{4\pi A}{h^2 c^2} \int_0^{\varepsilon_F} \varepsilon^2 d\varepsilon \end{aligned}$$

(c) Evaluate the integrals over ε .

$$\begin{aligned} \int_0^{\varepsilon_F} d\varepsilon &= \frac{\varepsilon^2}{2} \Big|_0^{\varepsilon_F} = \frac{1}{2} \varepsilon_F^2 \\ \int_0^{\varepsilon_F} \varepsilon^2 d\varepsilon &= \frac{\varepsilon^3}{3} \Big|_0^{\varepsilon_F} = \frac{1}{3} \varepsilon_F^3 \end{aligned}$$

(d) Eliminate the chemical potential to get an expression for U as a function of N and A .

$$\begin{aligned} N &= \frac{2\pi A}{h^2 c^2} \varepsilon_F^2 \implies \varepsilon_F = \left(\frac{h^2 c^2}{2\pi} \frac{N}{A} \right)^{1/2} = hc \sqrt{\frac{N}{2\pi A}} \\ U &= \frac{4\pi A}{3h^2 c^2} \varepsilon_F^3 = \frac{4\pi A}{3h^2 c^2} \left(hc \sqrt{\frac{N}{2\pi A}} \right)^3 \\ &= \frac{2}{3} N hc \sqrt{\frac{N}{2\pi A}} = \frac{2}{3} N \varepsilon_F \end{aligned}$$