

# Ideal Bose Gas

single-particle quantum states:  $i$

orthonormal:  $\langle i | i' \rangle = \delta_{ii'}$

complete  $\sum_i |i\rangle \langle i| = \mathbb{1}$

energy eigenstate:  $H_1 |i\rangle = \epsilon_i |i\rangle$

microstates for identical bosons

specify by occupation number  $n_i$

for each single-particle state  $i$

occupation numbers:  $n_i = 0, 1, 2, \dots$

number of bosons:  $N = \sum_i n_i$

↑ sum over single particle states

total energy:  $E = \sum_i n_i \epsilon_i$

Example: simple atoms in box

$$|i\rangle = |\vec{p}\rangle \quad \sum_i = \frac{1}{(2\pi\hbar)^3} \int d^3r \int d^3p$$

Grand Canonical Ensemble with temperature  $T = 1/\beta$   
chemical potential  $\mu$

grand partition function:

$$Q = e^{-\beta\Phi} = \prod_i \frac{1}{1 - e^{-\beta(\epsilon_i - \mu)}}$$

↖ product over single-particle states

grand potential:  $\Phi = U - ST - N\mu$

$$\log Q = -\beta\Phi = \sum_i \log \frac{1}{1 - e^{-\beta(\epsilon_i - \mu)}}$$

↖ sum over single-particle states

average occupation number:  $\langle n_i \rangle = \frac{1}{e^{\beta(\epsilon_i - \mu)} - 1}$

average particle number:  $\langle N \rangle = \sum_i \langle n_i \rangle = \sum_i \frac{1}{e^{\beta(\epsilon_i - \mu)} - 1}$

average energy:  $U = \sum_i \langle n_i \rangle \epsilon_i = \sum_i \frac{\epsilon_i}{e^{\beta(\epsilon_i - \mu)} - 1}$

thermodynamic relation:  $d\Phi = -SdT - PdV - Nd\mu$

entropy:  $S = -\left(\frac{\partial\Phi}{\partial T}\right)_{V,\mu} = +\frac{\partial}{\partial T}(T \log Q)_{V,\mu}$

pressure:  $P = -\left(\frac{\partial\Phi}{\partial V}\right)_{T,\mu} = +T \frac{\partial}{\partial V}(\log Q)_{T,\mu}$

simple bosonic atoms in volume V

single-particle energies are determined by momentum  $\vec{p}$

$$E(\vec{p}) = \frac{1}{2m} \vec{p}^2$$

sum over single-particle states

$$\begin{aligned} \sum_i &= \frac{1}{(2\pi\hbar)^3} \int d^3r \int d^3p \\ &= \frac{1}{(2\pi\hbar)^3} V 4\pi \int_0^\infty p^2 dp \\ &= \frac{1}{(2\pi\hbar)^3} V 2\pi (2m)^{3/2} \int_0^\infty d\epsilon \epsilon^{1/2} \end{aligned}$$

(average) number of particles

$$\begin{aligned} N &= \sum_i \frac{1}{e^{\beta(\epsilon_i - \mu)} - 1} \\ &= \frac{2\pi V}{(2\pi\hbar)^3} (2m)^{3/2} \int_0^\infty d\epsilon \epsilon^{1/2} \frac{1}{e^{\beta(\epsilon - \mu)} - 1} \\ &= \frac{2\pi V}{(2\pi\hbar)^3} (2m)^{3/2} \frac{1}{\beta^{3/2}} \int_0^\infty dx \frac{x^{1/2}}{e^{-\beta\mu} e^x - 1} \quad x = \beta\epsilon \\ &= \frac{V}{\lambda_{th}^3} g_{3/2}(e^{\beta\mu}) \quad \lambda_{th} = \sqrt{\frac{2\pi\hbar^2}{m k T}} \end{aligned}$$

Bose-Einstein function:  $g_\nu(z) = \frac{1}{\Gamma(\nu)} \int_0^\infty dx \frac{z^{\nu-1}}{e^x - 1}$

Thermodynamics parameter in  $Z = e^{\beta P}$

$$N = \frac{V}{\lambda_{th}^3} g_{3/2}(z) \quad \lambda_{th} = \sqrt{\frac{2\pi\hbar^2}{m k T}}$$

$$U = \frac{3}{2} k T \frac{V}{\lambda_{th}^3} g_{5/2}(z)$$

$$P = \frac{k T}{\lambda_{th}^3} g_{5/2}(z)$$

$$S = \frac{V}{\lambda_{th}^3} \left[ \frac{5}{2} g_{5/2}(z) - g_{3/2}(z) \log z \right]$$

Simple relation at all temperature:

$$U = \frac{3}{2} P V$$

high-temperature limit:  $\beta \rightarrow 0$ ,  $z \rightarrow \infty$ ,  $g_\nu(z) \rightarrow z$

$$P V = N k T$$

low-temperature limit:  $\beta \rightarrow \infty$ ,  $z \rightarrow 1$ ,  $g_\nu(z) \rightarrow \zeta_\nu$

$$N = \zeta_{3/2} \frac{V}{\lambda_{th}^3} \quad P = \zeta_{5/2} \frac{k T}{\lambda_{th}^3} \quad \zeta_{3/2} = 2.612$$

$$\zeta_{5/2} = 1.341$$

$$P V = \frac{\zeta_{5/2}}{\zeta_{3/2}} N k T = 0.5135 N k T$$

Problem at low temperature:

$$\frac{N}{V} \left( \frac{2\pi\hbar^2}{mkT} \right)^{3/2} < \frac{1}{5^{3/2}}$$

lower bound on temperature:  $kT > \frac{2\pi\hbar^2}{m} \left( \frac{1}{5^{3/2}} \frac{N}{V} \right)^{2/3}$

What happens if you decrease  $T$  further?

Einstein's solution: Bose-Einstein condensation

macroscopic number  $N_0$  of particles

in lowest-energy quantum state

energy =  $\epsilon_0$

must take that state into account explicitly,

sum over all other single-particle states

can be expressed as integral over  $\vec{r}$  and  $\vec{p}$

thermodynamic variables as function of  $z = e^{\beta\mu}$

$$N = \frac{1}{e^{\beta(\epsilon_0 - \mu)} - 1} + \frac{V}{\lambda_{th}^3} g_{3/2}(z)$$

$$U = \frac{\epsilon_0}{e^{\beta(\epsilon_0 - \mu)} - 1} + \frac{3}{2} kT \frac{V}{\lambda_{th}^3} g_{5/2}(z)$$

$$P = \frac{1}{e^{\beta(\epsilon_0 - \mu)} - 1} \left( -\frac{\partial \epsilon_0}{\partial V} \right) + \frac{kT}{\lambda_{th}^3} g_{5/2}(z)$$

$$S = \frac{PV + U}{kT} - (\log z) N$$

lowest energy state in cubic box of length  $L$

periodic boundary condition:  $\epsilon_0 = 0$

vanishing boundary condition:  $\epsilon_0 = \frac{1}{2m} \left( \frac{2\pi\hbar}{L} \right)^2 (1+1+1)$

$$= 6\pi^2 \frac{\hbar^2}{m V^{2/3}}$$

$$-\frac{\partial \epsilon_0}{\partial V} = +\frac{2}{3} \frac{\epsilon_0}{V}$$

thermodynamic limit:  $N \rightarrow \infty, V \rightarrow \infty, \frac{N}{V}$  fixed

If  $N_0 = \frac{1}{e^{\beta(\epsilon_0 - \mu)} - 1}$  has thermodynamic limit

$$\frac{\epsilon_0}{e^{\beta(\epsilon_0 - \mu)} - 1} \sim \frac{N_0}{V^{2/3}}$$

is also,  $\frac{1}{e^{\beta(\epsilon_0 - \mu)} - 1} \left( -\frac{\partial \epsilon_0}{\partial V} \right) \sim \frac{N_0}{V^{5/3}}$

Simplified treatment in thermodynamic limit

$T > T_c$ : ignore lowest energy state

$$\mu < 0 \text{ in } g_r(e^{\beta\mu})$$

$T \leq T_c$ : keep lowest energy state in  $N$  but not in  $M, P, S$

$$\mu = 0 \text{ in } g_r(e^{\beta\mu}): g_r(1) = S_r$$

treat  $(T, N_0)$  as parameters instead of  $(T, \mu)$

thermodynamic variables below  $T_c$

$$N = N_0(T) + S_{3/2} \frac{V}{\lambda_{Th}^3}$$

$$U = 0 + S_{5/2} \frac{3}{2} kT \frac{V}{\lambda_{Th}^3}$$

$$P = 0 + S_{5/2} \frac{kT}{\lambda_{Th}^3}$$

$$S = 0 + S_{5/2} \cdot \frac{5}{2} \frac{V}{\lambda_{Th}^3}$$

critical temperature:  $T_c$   $N_0(T_c) = 0$

$$N = 0 + S_{3/2} V \left( \frac{mkT_c}{2\pi\hbar^2} \right)^{3/2}$$

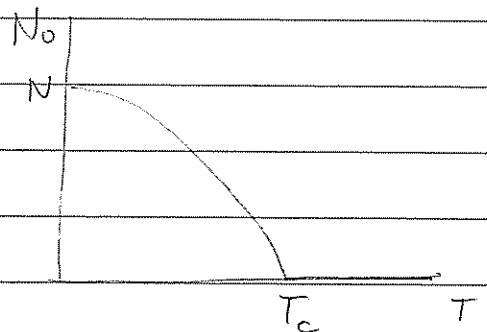
$$\implies kT_c = \frac{2\pi\hbar^2}{m} \left( \frac{1}{S_{3/2}} \frac{N}{V} \right)^{2/3}$$

eliminate  $V$  in equation for  $N$

$$N = N_0(T) + N \left( \frac{T}{T_c} \right)^{3/2}$$

solve for  $N_0(T)$ :

$$N_0(T) = N \left[ 1 - \left( \frac{T}{T_c} \right)^{3/2} \right]$$



The decrease in temperature from  $T_c$  to 0  
is a phase transition

from a Bose gas with 0 chemical potential

$$PV \approx 0.5135 NkT$$

$$U \approx 0.5135 \cdot \frac{3}{2} NkT$$

$$S \approx 0.5135 \cdot \frac{5}{2} N$$

to a condensed phase in which all the bosons  
are in the same lowest energy quantum state

$$P=0, U=0, S=0$$

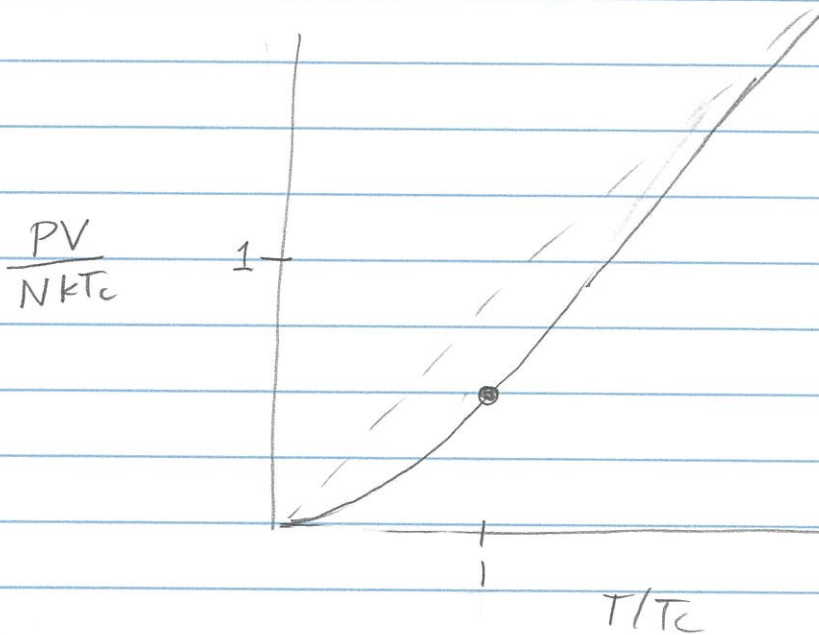
The region  $0 < T < T_c$  is a coexistence region  
between the two phases

As  $T$  decreases from  $T_c$  to 0,

the fraction of atoms in the condensed phase  
increases from 0 to 1



# Pressure vs. temperature



At  $T = T_c$ ,  $P, U, S$  are all continuous  
and their first derivative are continuous  
but they have 2<sup>nd</sup> derivative that are discontinuous

"2<sup>nd</sup> order phase transition"

specific heat at constant volume

