

# Ensembles

## Macroscopic system

example: gas of  $N$  atoms confined to volume  $V$

$$N \gg 1 \quad \log N \gg 1$$

$$V \gg \text{volume } v_0 \text{ of atom} \quad \log \frac{V}{v_0} \gg 1$$

macrostate characterized by macroscopic properties

equilibrium: macroscopic properties don't change  
with time

macrostate for gas in equilibrium  
is characterized by 3 variables:

$N$ ,  $V$ , and energy  $E$   
or temperature  $T$

microstates for classical gas

specified by position vector:  $\vec{r}_n = (x_n, y_n, z_n)$

and momentum vector  $\vec{p}_n = (p_{nx}, p_{ny}, p_{nz})$

for each of the atoms  $n = 1, \dots, N$

OR by  $3N$  Cartesian coordinates:  $x_1, x_2, \dots, x_{3N}$

and  $3N$  momentum components:  $p_1, p_2, \dots, p_{3N}$

$\implies 6N$  continuous variables

microstate for quantum gas

specified by independent solutions

of Schrödinger equation

$$\hat{H} |\psi_i\rangle = E_i |\psi_i\rangle$$

finite volume  $\implies$  energy eigenvalues  $E_i$  are discrete

$\implies$  microstates can be enumerated

$$|\psi_i\rangle, \quad i = 1, 2, 3, \dots$$

ensemble: probability distribution  
for microstates

quantum system: microstates can be enumerated

probability distribution is specified by  
the probability  $P_i$  for each microstate

$$P_i \geq 0 \quad \text{for each } i$$

$$\sum_i P_i = 1$$

classical system

generalized coordinates:  $q_1, q_2, \dots, q_N$

conjugate momenta:  $p_1, p_2, \dots, p_N$

integration measure on phase space

$$d\omega = dp_1 dq_1 dp_2 dq_2 \dots dp_N dq_N$$

probability distribution is a function  $P(p, q)$

$$P(p, q) \geq 0$$

$$\int d\omega P(p, q) = 1$$

# Statistical Mechanics

derive thermodynamic variables  
from probability distribution of ensemble

information entropy of ensemble

for discrete microstates

$$S_{\text{info}} = - \sum_i P_i \log P_i$$

for classical system.

$$S_{\text{info}} = - \int dw P(p, q) \log P(p, q)$$

all probability in one microstate:  $P_1 = 1$

$P_i = 0$  for  $i = 2, 3, \dots$

$$\implies S_{\text{info}} = 0$$

equal probability in all  $N$  microstate:  $P_i = \frac{1}{N}$

$$\implies S_{\text{info}} = \log N$$

energy  $E$  of macrostate

microstate  $i$  has discrete energy  $E_i$

average over ensemble:

$$\langle E \rangle = \sum_i P_i E_i$$

classical system

microstate  $(p, q) = (p_1, p_2, \dots, p_N; q_1, q_2, \dots, q_N)$

has energy  $\mathcal{H}(p, q)$

$$\langle E \rangle = \int dw P(p, q) \mathcal{H}(p, q)$$