

# Grand Canonical Ensemble

macroscopic system with energy  $E$   
number of particles  $N$

\* of microstates:  $\Omega(E, N)$

much larger system (reservoir) with energy  $E_{res}$   
number of particles  $N_{res}$

# of microstates:  $\Omega_{res}(E_{res}, N_{res})$

system and reservoir

in thermal contact: can exchange energy  
but total energy  $E_{tot}$  is fixed  
if system has energy  $E$ , reservoir has energy  $E_{tot} - E$

in chemical contact: can exchange particles  
but total number of particles  $N_{tot}$  is fixed  
if system has  $N$  particles, reservoir has  $N_{tot} - N$  particles

if system has energy  $E$  and particle number  $N$

# of accessible microstates  $\approx \Omega(E, N) \times \Omega_{res}(E_{tot} - E, N_{tot} - N)$

reservoir much larger than system

$$\Rightarrow E \ll E_{\text{tot}}, N \ll N_{\text{tot}}$$

expand  $\log \Omega_{\text{res}}$  to 1<sup>st</sup> order in  $E$  and  $N$

$$\log \Omega_{\text{res}}(E_{\text{tot}} - E, N_{\text{tot}} - N) = S_{\text{res}}(E_{\text{tot}} - E, N_{\text{tot}} - N)$$

$$\approx S_{\text{res}}(E_{\text{tot}}, N_{\text{tot}}) + \frac{\partial S_{\text{res}}}{\partial E_{\text{res}}}(E_{\text{tot}}, N_{\text{tot}})(-E) + \frac{\partial S_{\text{res}}}{\partial N_{\text{res}}}(E_{\text{tot}}, N_{\text{tot}})(-N)$$

$$= S_{\text{res}}(E_{\text{tot}}, N_{\text{tot}}) - \frac{1}{T_{\text{res}}} E + \frac{M_{\text{res}}}{T_{\text{res}}} N$$

# of accessible microstates:

$$\Omega(E, N) \times \exp\left(S_{\text{res}}(E_{\text{tot}}, N_{\text{tot}}) - \frac{1}{T_{\text{res}}} E + \frac{M_{\text{res}}}{T_{\text{res}}} N\right)$$

$$= \Omega_{\text{res}}(E_{\text{tot}}, N_{\text{tot}}) \times \Omega(E, N) \exp\left(-\frac{1}{T_{\text{res}}} E + \frac{M_{\text{res}}}{T_{\text{res}}} N\right)$$

probability distribution for microstates of system

$$P(E_r, N_r) = \frac{1}{Z} \exp(-\beta E_r - \alpha N_r)$$

$$\beta = \frac{1}{T_{\text{res}}} \quad \alpha = -\frac{M_{\text{res}}}{T_{\text{res}}}$$

$$Z = \sum_r \exp(-\beta E_r - \alpha N_r)$$

grand canonical ensemble with temperature  $T$   
chemical potential  $\mu$

probability distribution of microstates  $i$

$$P_i = \frac{1}{\mathcal{Z}} \exp(-\beta E_i - \alpha N_i) \quad \beta = \frac{1}{T} \quad \alpha = -\frac{\mu}{T}$$

$$\mathcal{Z} = \sum_r \exp(-\beta E_r - \alpha N_r) \Rightarrow \sum_r P_r = 1$$

grand partition function:  $\mathcal{Z}(\beta, \alpha)$

average energy:  $U = \langle E_r \rangle = \frac{1}{\mathcal{Z}} \sum_r E_r \exp(-\beta E_r - \alpha N_r)$

$$= -\frac{\partial}{\partial \beta} \log \mathcal{Z}(\beta, \alpha)$$

average number:  $\bar{N} = \langle N_i \rangle = \frac{1}{\mathcal{Z}} \sum_r N_r \exp(-\beta E_r - \alpha N_r)$

$$= -\frac{\partial}{\partial \alpha} \log \mathcal{Z}(\beta, \alpha)$$

information entropy:

$$S = -\sum_r P_r \log P_r = -\sum_r \left( \frac{1}{\mathcal{Z}} e^{-\beta E_r - \alpha N_r} \right) (-\beta E_r - \alpha N_r - \log \mathcal{Z})$$
$$= \beta \frac{1}{\mathcal{Z}} \sum_r E_r e^{-\beta E_r - \alpha N_r} + \alpha \frac{1}{\mathcal{Z}} \sum_r N_r e^{-\beta E_r - \alpha N_r} + \frac{\log \mathcal{Z}}{\mathcal{Z}} \sum_r e^{-\beta E_r - \alpha N_r}$$
$$= \beta U + \alpha \bar{N} + \log \mathcal{Z}$$

$$\log Z = -\beta U - \alpha \bar{N} + S$$

define grand potential  $\Phi$  by  $Z = e^{-\beta\Phi}$

$$\log Z = -\beta\Phi$$

$$\Phi = U - TS - \mu \bar{N}$$

thermodynamic relation

$$\begin{aligned} d\Phi &= (TdS - PdV + \mu d\bar{N}) - (TdS + SdT) - (\mu d\bar{N} + \bar{N}d\mu) \\ &= -SdT - PdV - \bar{N}d\mu \end{aligned}$$

partial derivative:

$$S = -\left(\frac{\partial\Phi}{\partial T}\right)_{\mu, V}$$

$$P = -\left(\frac{\partial\Phi}{\partial V}\right)_{T, \mu}$$

$$\bar{N} = -\left(\frac{\partial\Phi}{\partial\mu}\right)_{T, V}$$

Gibbs free energy:  $G = U - TS + PV$

exact extensivity:  $G = \bar{N}\mu \implies \Phi = -PV$

grand partition function:

$$Z = \sum_r e^{-\beta E_r - \alpha N_r}$$

make sum over particle number  $N$  explicit

$$\begin{aligned} Z &= \sum_{N=0}^{\infty} \sum_{r: N_r=N} e^{-\beta E_r - \alpha N_r} \\ &= \sum_{N=0}^{\infty} e^{-\alpha N} \sum_{r: N_r=N} e^{-\beta E_r} \\ &= \sum_{N=0}^{\infty} (e^{-\alpha})^N Z_N(\beta) \end{aligned}$$

$Z_N(\beta) =$  partition function for  $N$  particles  
in canonical ensemble

fugacity:  $z = e^{-\alpha} = e^{\beta\mu}$

example: monatomic ideal gas

$$Z_N = \frac{1}{N!} \left( V \left( \frac{m}{2\pi k^2 \beta} \right)^{3/2} \right)^N$$

$$Z = \sum_{N=0}^{\infty} z^N \frac{1}{N!} \left( V \left( \frac{m}{2\pi k^2 \beta} \right)^{3/2} \right)^N$$

$$= \exp \left( z V \left( \frac{m}{2\pi k^2 \beta} \right)^{3/2} \right) = \exp \left( e^{\mu/T} V \left( \frac{mT}{2\pi k^2} \right)^{3/2} \right)$$

$$\log Z = e^{\mu/T} V \left( \frac{mT}{2\pi k^2} \right)^{3/2}$$

fluctuation in the grand canonical ensemble

mean-square deviations

$$\begin{aligned} \text{in energy: } (\Delta E)^2 &\equiv \langle (E - U)^2 \rangle = \langle E^2 \rangle - U^2 \\ &= - \left( \frac{\partial U}{\partial \beta} \right)_{\mu, V} = + T^2 \left( \frac{\partial U}{\partial T} \right)_{\mu, T, V} \end{aligned}$$

$$\begin{aligned} \text{in number } (\Delta N)^2 &\equiv \langle (N - \bar{N})^2 \rangle = \langle N^2 \rangle - \bar{N}^2 \\ &= - \left( \frac{\partial \bar{N}}{\partial \mu} \right)_{\beta, V} = + T \left( \frac{\partial \bar{N}}{\partial \mu} \right)_{T, V} \end{aligned}$$

$U$  and  $\bar{N}$  are extensive, scale as  $N^1$   
 $\Delta E$  and  $\Delta N$  scale as  $N^{1/2}$

relative size of fluctuation  $\frac{\Delta E}{U}, \frac{\Delta N}{\bar{N}}$  scale as  $N^{-1/2}$   
go to 0 in thermodynamic limit

fluctuations in number density  $n = N/V$

$$\begin{aligned} \frac{(\Delta n)^2}{n^2} &= \frac{T}{\bar{N}^2} \left( \frac{\partial \bar{N}}{\partial \mu} \right)_{T, V} \\ &= \frac{T}{\bar{N}^2} \left( - \frac{V}{v^2} \frac{\partial v}{\partial \mu} \right)_{T, V} \\ &= - \frac{T}{V} \left( \frac{1}{v} \frac{\partial v}{\partial \mu} \right)_{T, V} = \frac{T}{V} K_T \\ &\sim N^{-1} \text{ unless } K_T = \infty \end{aligned}$$

$\bar{N} = V \cdot \frac{N}{V} = \frac{V}{v}$  where  $v = \frac{V}{N}$   
 $dp = v dp - s dT$   
 $K_T = \text{isothermal compressibility}$