

Supporting Information

Theoretical approach to evaluate the gas-sensing performance of graphene nanoribbon/oligothiophene composites

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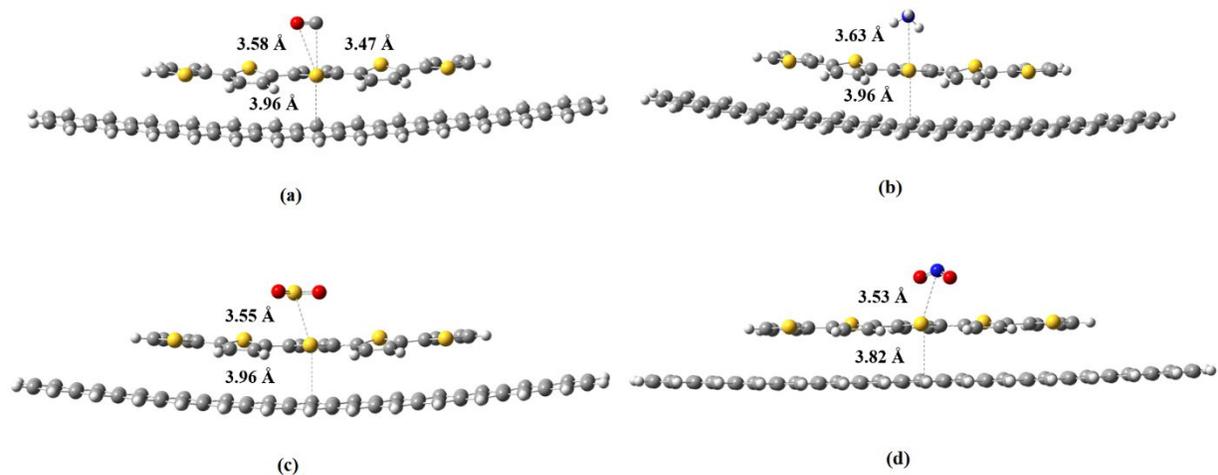


Figure S1: Optimized geometries of (a) C₅₄H₃₀ ... 5PT ... CO, (b) C₅₄H₃₀ ... 5PT ... NH₃, (c) C₅₄H₃₀ ... 5PT ... SO₂ and (d) C₅₄H₃₀ ... 5PT ... NO₂ composite-analyte complexes

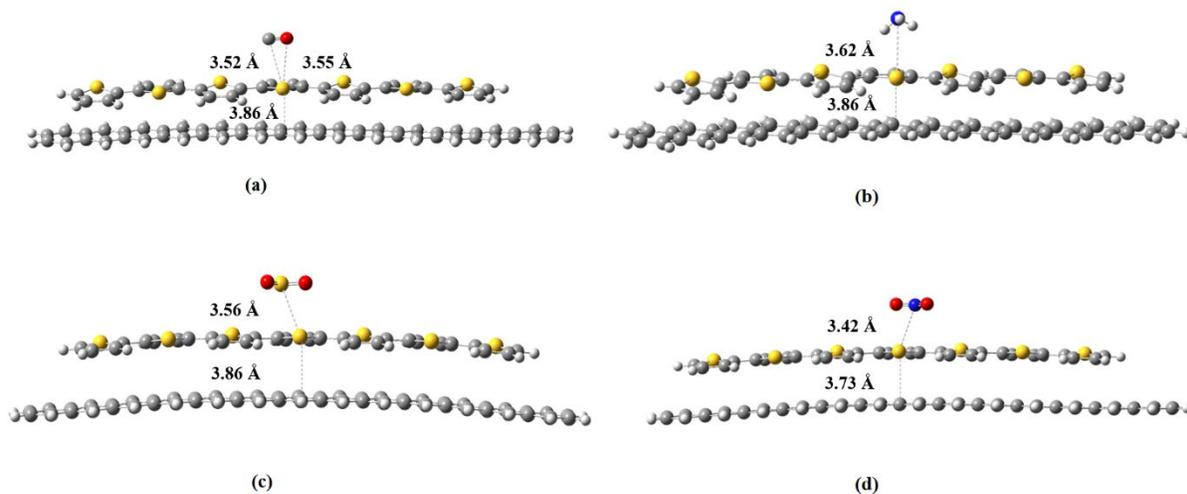


Figure S2: Optimized geometries of (a) C₅₄H₃₀ ... 7PT ... CO, (b) C₅₄H₃₀ ... 7PT ... NH₃, (c) C₅₄H₃₀ ... 7PT ... SO₂ and (d) C₅₄H₃₀ ... 7PT ... NO₂ composite-analyte complexes

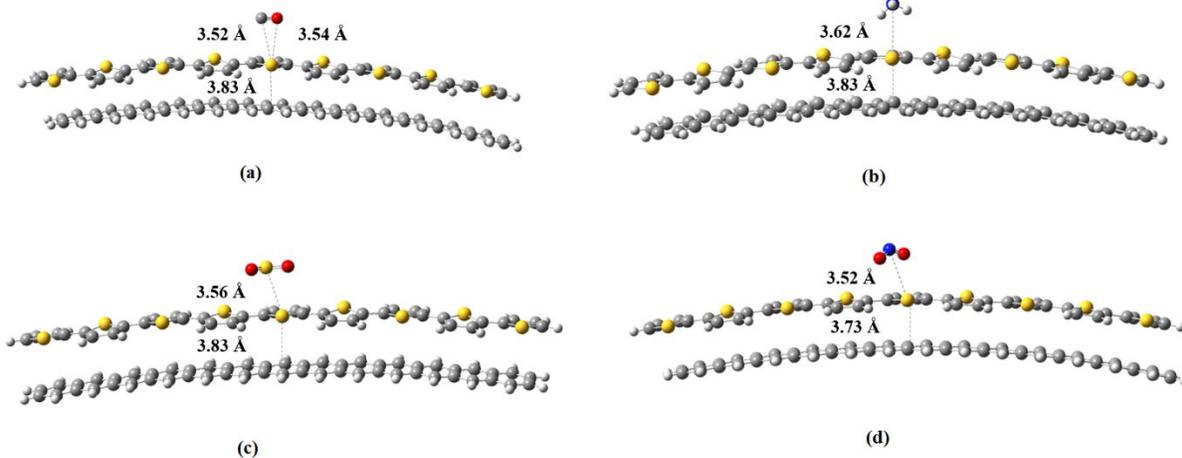


Figure S3: Optimized geometries of (a) $C_{54}H_{30} \dots 9PT \dots CO$, (b) $C_{54}H_{30} \dots 9PT \dots NH_3$, (c) $C_{54}H_{30} \dots 9PT \dots SO_2$ and (d) $C_{54}H_{30} \dots 9PT \dots NO_2$ composite-analyte complexes

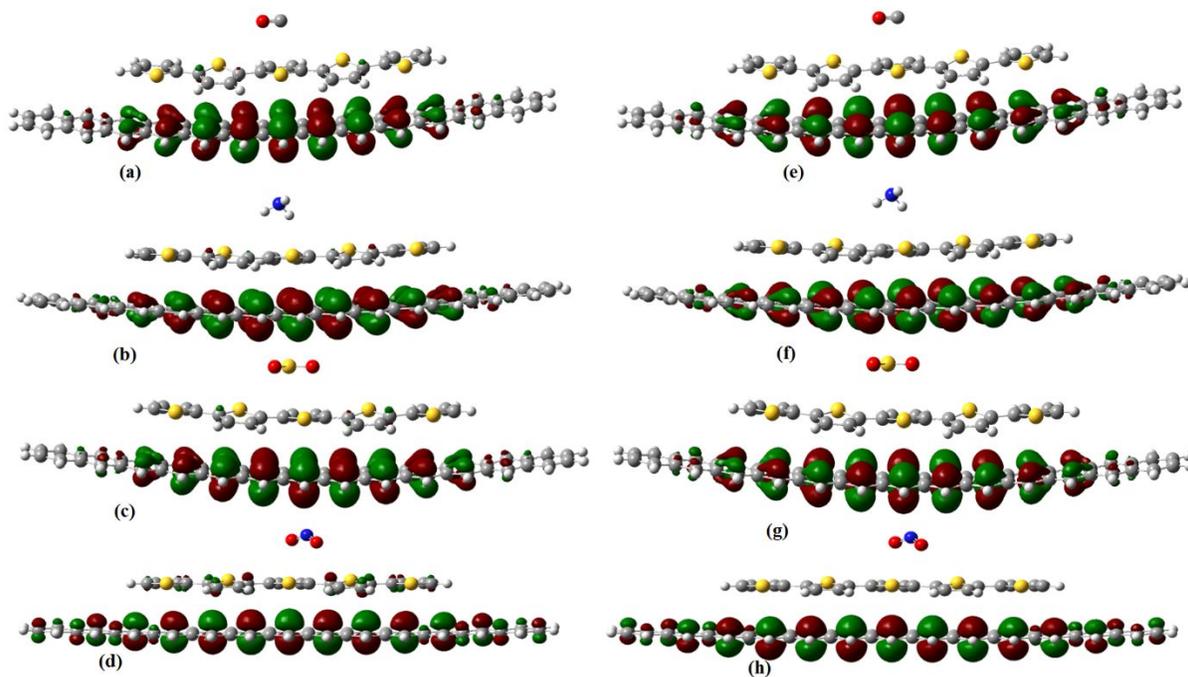


Figure S4: HOMO and LUMO orbitals of (a, e) $C_{54}H_{30} \dots 5PT \dots CO$, (b, f) $C_{54}H_{30} \dots 5PT \dots NH_3$, (c, g) $C_{54}H_{30} \dots 5PT \dots SO_2$ and (d, h) $C_{54}H_{30} \dots 5PT \dots NO_2$ composite-analyte complexes

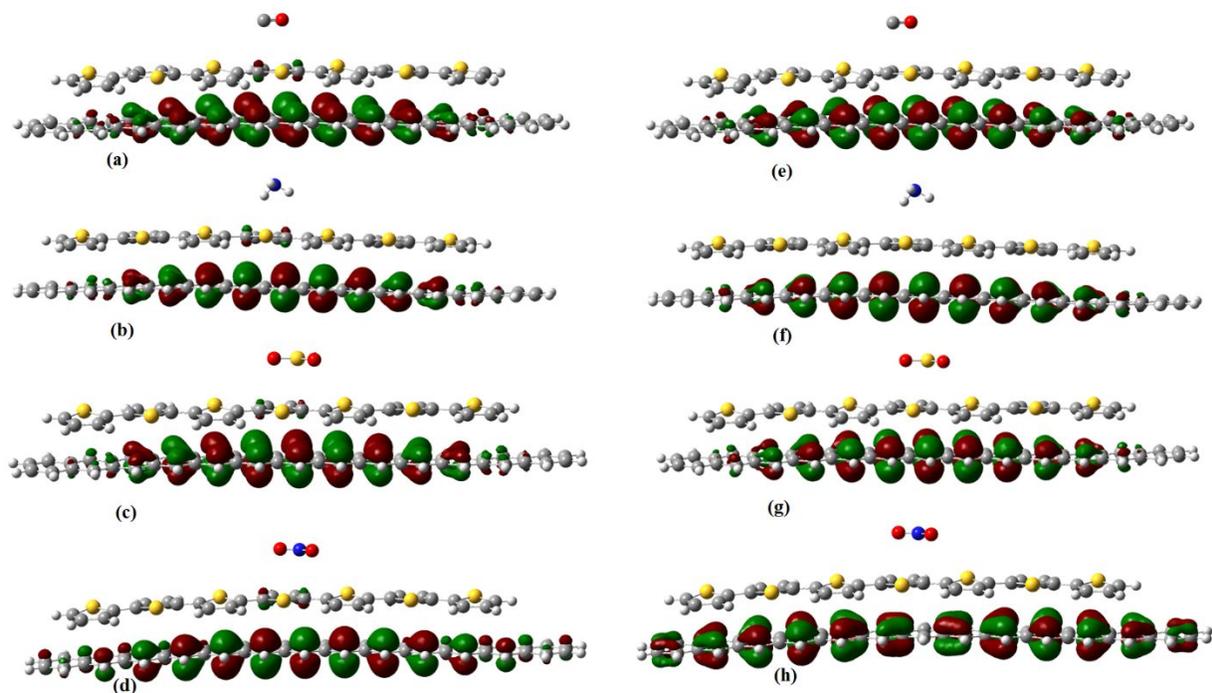


Figure S5: HOMO and LUMO orbitals of (a, e) C₅₄H₃₀ ... 7PT ... CO, (b, f) C₅₄H₃₀ ... 7PT ... NH₃, (c, g) C₅₄H₃₀ ... 7PT ... SO₂ and (d, h) C₅₄H₃₀ ... 7PT ... NO₂ composite-analyte complexes

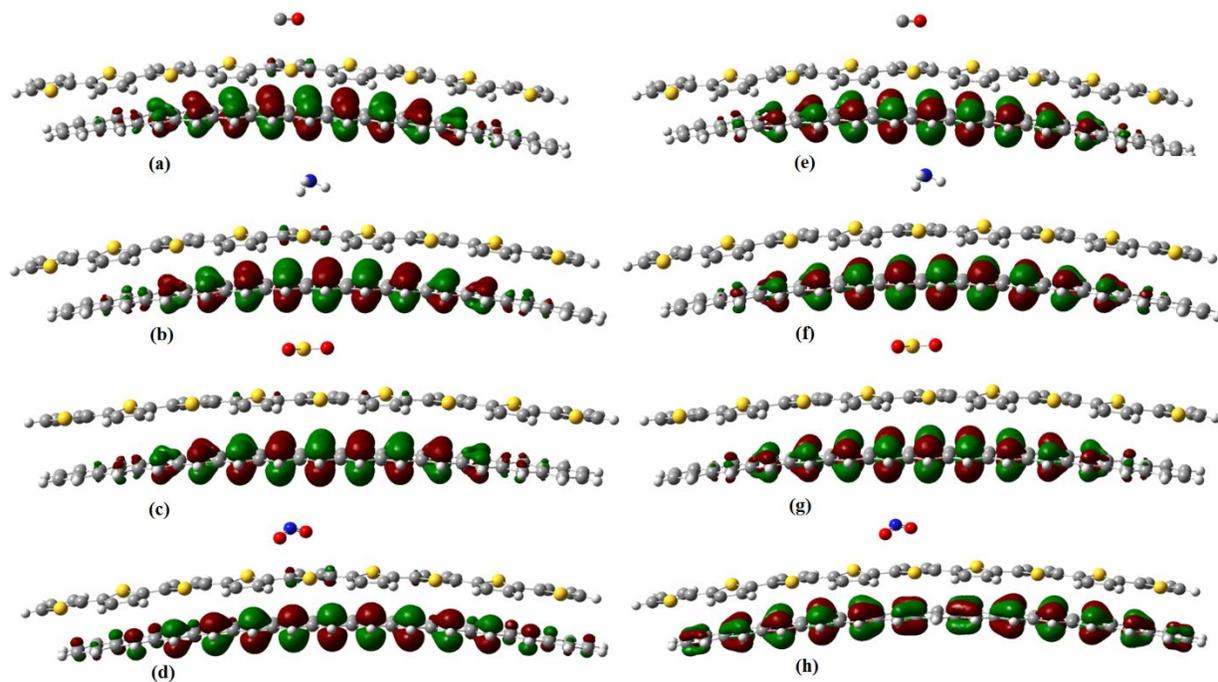


Figure S6: HOMO and LUMO orbitals of (a, e) C₅₄H₃₀ ... 9PT ... CO, (b, f) C₅₄H₃₀ ... 9PT ... NH₃, (c, g) C₅₄H₃₀ ... 9PT ... SO₂ and (d, h) C₅₄H₃₀ ... 9PT ... NO₂ composite-analyte complexes

Table S1: Ionization energy, electron affinity, chemical potential (μ), hardness (η), softness (S), and electrophilicity (ω) of $C_{54}H_{30} \dots nPT \dots CO$, $C_{54}H_{30} \dots nPT \dots NH_3$, $C_{54}H_{30} \dots nPT \dots SO_2$ and $C_{54}H_{30} \dots nPT \dots NO_2$ ($n= 3, 5, 7, 9$) composite-analyte complexes

System	I.E	E.A	Chemical Potential (μ)	Softness (S)	Hardness (η)	Electrophilicity (ω)
$C_{54}H_{30} \dots 3PT$	3.86	3.22	-3.54	1.58	0.32	19.80
$C_{54}H_{30} \dots 3PT \dots CO$	3.88	3.23	-3.55	1.54	0.32	19.47
$C_{54}H_{30} \dots 3PT \dots NH_3$	3.92	3.27	-3.59	1.54	0.32	19.92
$C_{54}H_{30} \dots 3PT \dots SO_2$	3.92	3.29	-3.61	1.60	0.31	20.85
$C_{54}H_{30} \dots 3PT \dots NO_2$	4.45	2.56	-3.51	0.53	0.94	6.51
$C_{54}H_{30} \dots 5PT$	3.78	3.19	-3.48	1.71	0.29	20.67
$C_{54}H_{30} \dots 5PT \dots CO$	3.79	3.19	-3.49	1.69	0.30	20.62
$C_{54}H_{30} \dots 5PT \dots NH_3$	3.83	3.24	-3.53	1.69	0.30	21.14
$C_{54}H_{30} \dots 5PT \dots SO_2$	3.82	3.23	-3.53	1.91	0.30	21.04
$C_{54}H_{30} \dots 5PT \dots NO_2$	4.28	2.62	-3.45	0.60	0.83	7.18
$C_{54}H_{30} \dots 7PT$	3.77	3.19	-3.48	1.72	0.29	20.77
$C_{54}H_{30} \dots 7PT \dots CO$	3.78	3.19	-3.49	1.72	0.29	20.86
$C_{54}H_{30} \dots 7PT \dots NH_3$	3.81	3.23	-3.52	1.72	0.29	21.26
$C_{54}H_{30} \dots 7PT \dots SO_2$	3.82	3.24	-3.53	1.72	0.29	21.42
$C_{54}H_{30} \dots 7PT \dots NO_2$	4.03	3.19	-3.61	1.19	0.42	15.53
$C_{54}H_{30} \dots 9PT$	3.76	3.18	-3.47	1.73	0.29	20.79
$C_{54}H_{30} \dots 9PT \dots CO$	3.77	3.19	-3.48	1.73	0.29	20.91
$C_{54}H_{30} \dots 9PT \dots NH_3$	3.81	3.22	-3.52	1.72	0.29	21.28
$C_{54}H_{30} \dots 9PT \dots SO_2$	3.80	3.21	-3.51	1.72	0.29	21.16
$C_{54}H_{30} \dots 9PT \dots NO_2$	4.01	3.39	-3.70	1.63	0.31	22.38

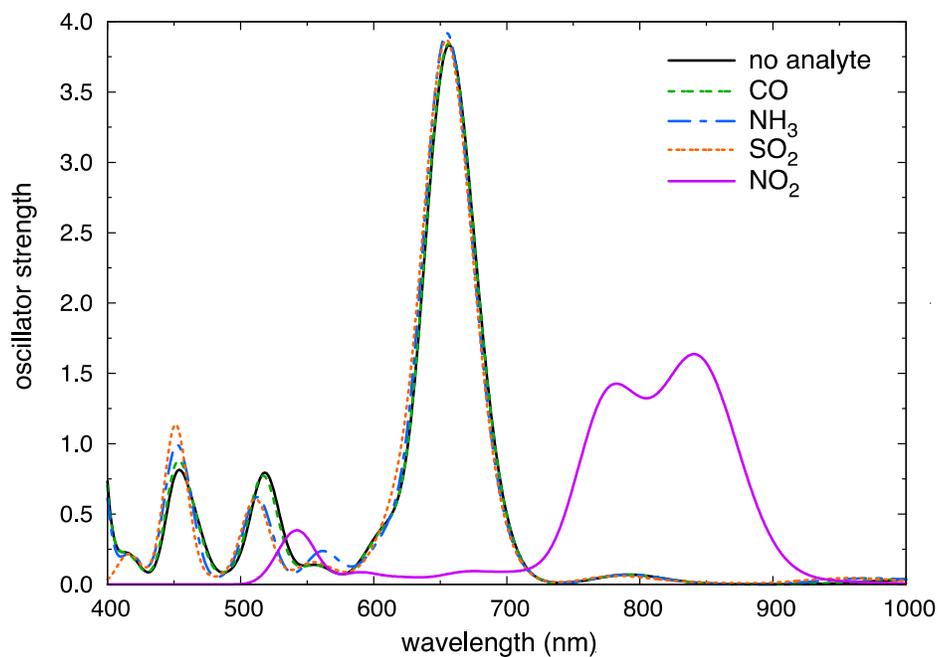


Figure S7: UV-vis spectra of $C_{54}H_{30} \dots 5PT$, $C_{54}H_{30} \dots 5PT \dots CO$, $C_{54}H_{30} \dots 5PT \dots NH_3$, $C_{54}H_{30} \dots 5PT \dots SO_2$ and $C_{54}H_{30} \dots 5PT \dots NO_2$ composite-analyte complexes, computed at the TD-B3LYP/6-31G** level.

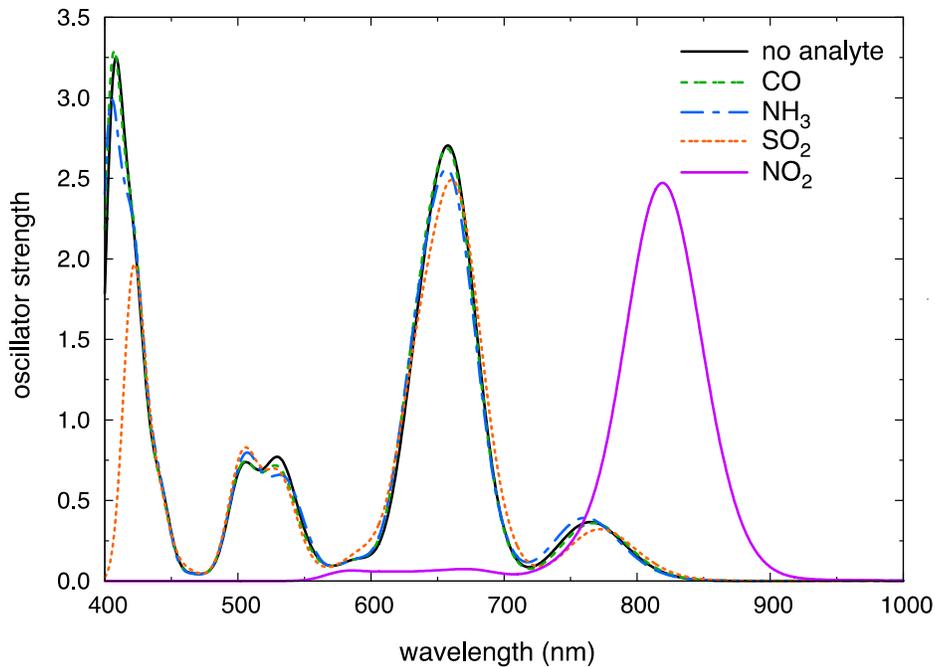


Figure S8: UV-vis spectra of $C_{54}H_{30} \dots 7PT$, $C_{54}H_{30} \dots 7PT \dots CO$, $C_{54}H_{30} \dots 7PT \dots NH_3$, $C_{54}H_{30} \dots 7PT \dots SO_2$ and $C_{54}H_{30} \dots 7PT \dots NO_2$ composite-analyte complexes, computed at the TD-B3LYP/6-31G** level.

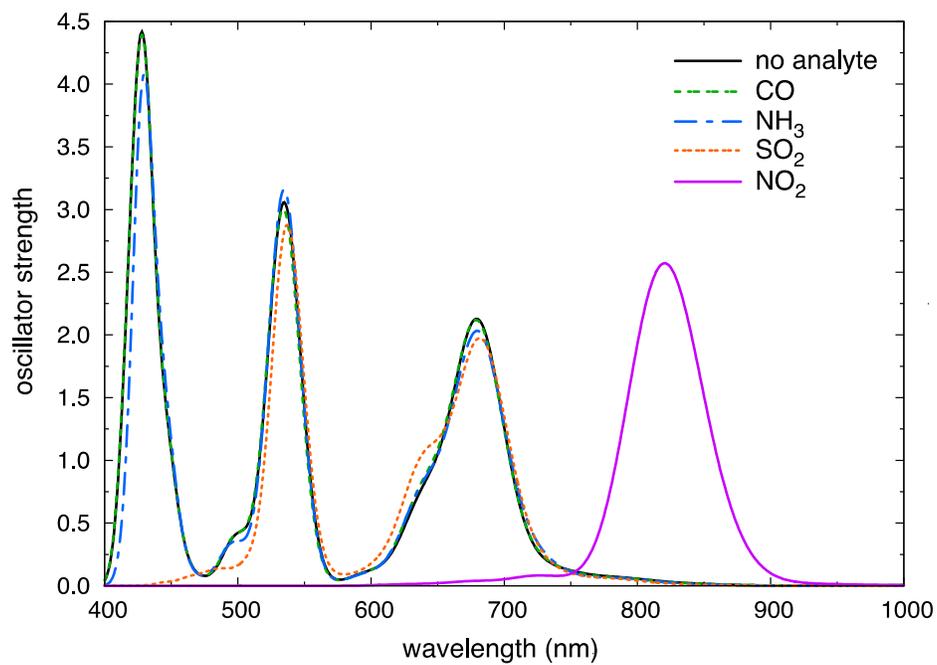


Figure S9: UV-vis spectra of C₅₄H₃₀... 9PT, C₅₄H₃₀... 9PT... CO, C₅₄H₃₀... 9PT... NH₃, C₅₄H₃₀... 9PT... SO₂ and C₅₄H₃₀... 9PT... NO₂ composite-analyte complexes, computed at the TD-B3LYP/6-31G** level.

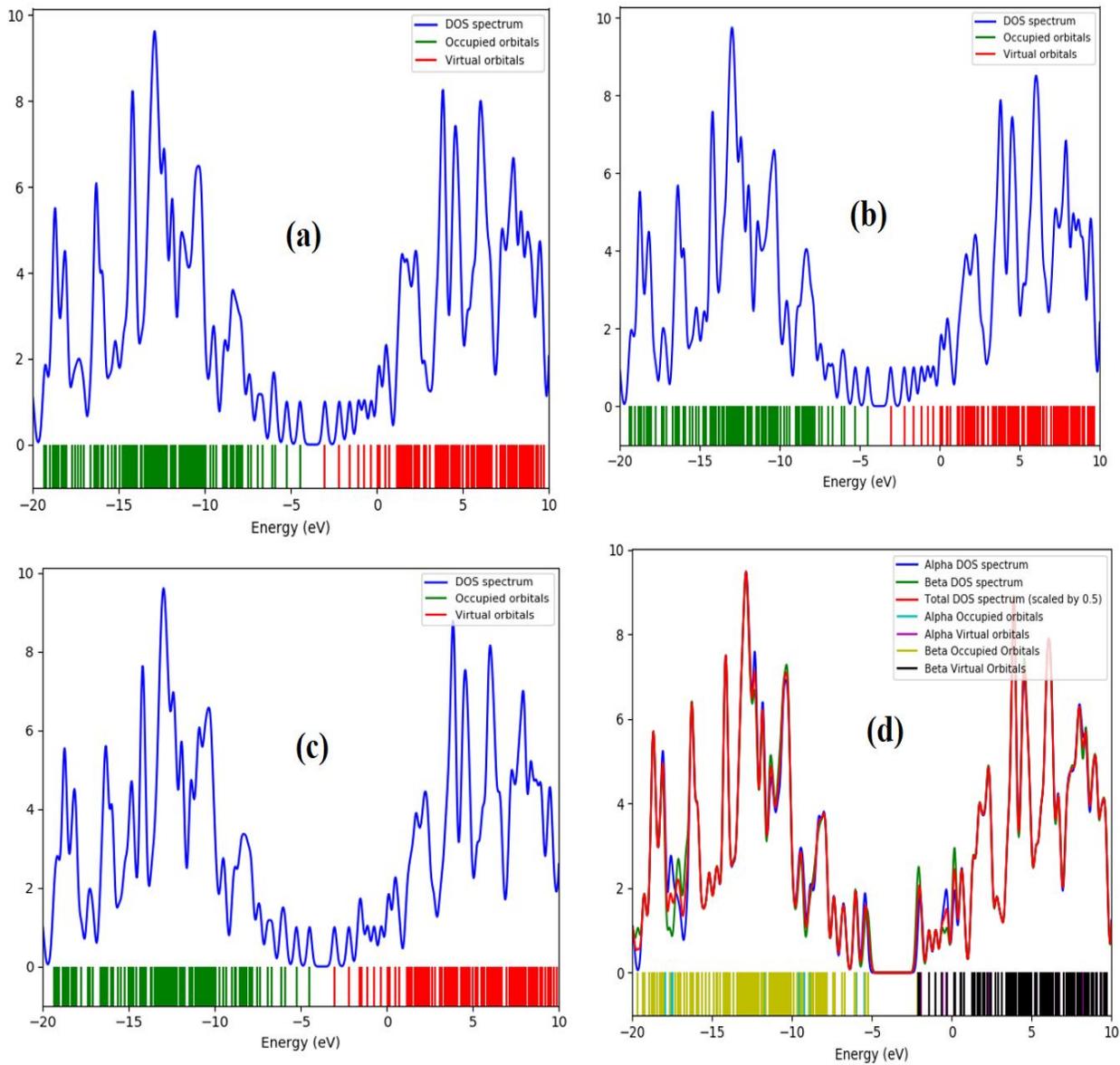


Figure S10: DOS spectra of (a) $C_{54}H_{30} \dots 5PT \dots CO$, (b) $C_{54}H_{30} \dots 5PT \dots NH_3$, (c) $C_{54}H_{30} \dots 5PT \dots SO_2$ and (d) $C_{54}H_{30} \dots 5PT \dots NO_2$ composite-analyte complexes

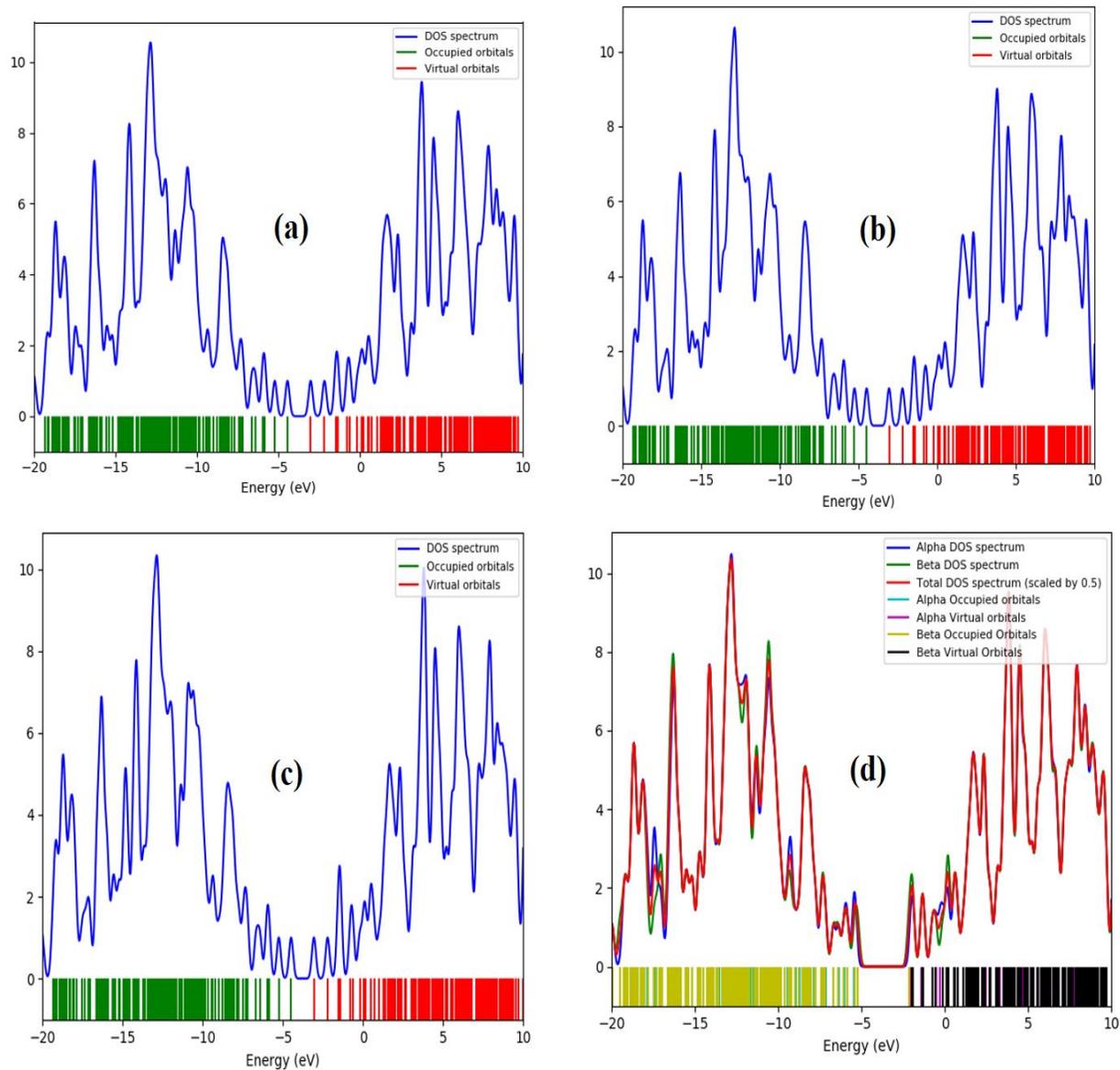


Figure S11: DOS spectra of (a) $C_{54}H_{30} \dots 7PT \dots CO$, (b) $C_{54}H_{30} \dots 7PT \dots NH_3$, (c) $C_{54}H_{30} \dots 7PT \dots SO_2$ and (d) $C_{54}H_{30} \dots 7PT \dots NO_2$ composite-analyte complexes

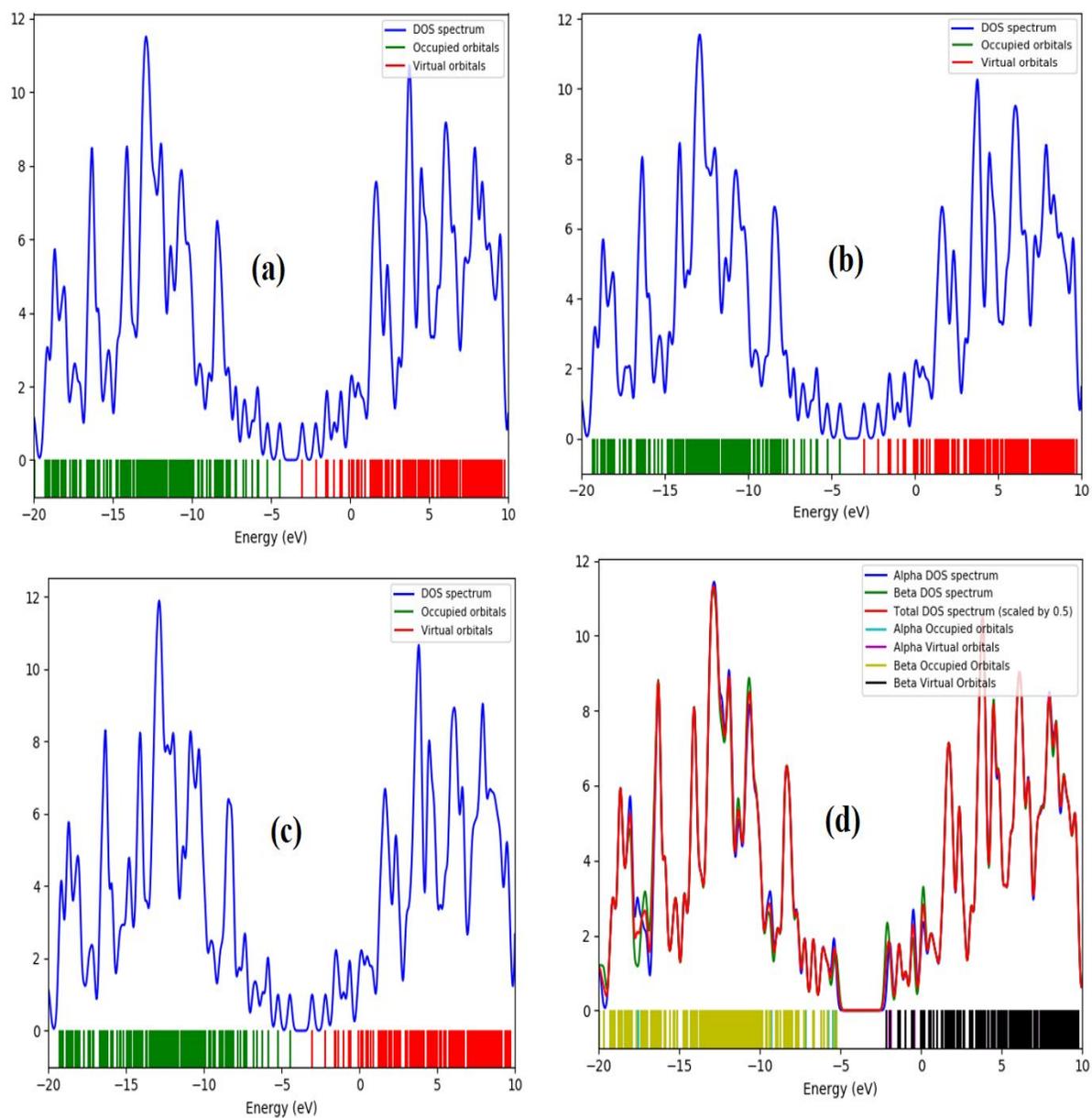
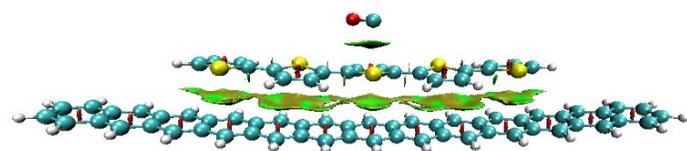
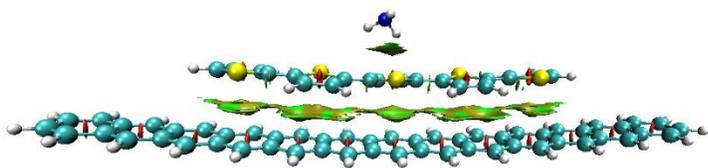
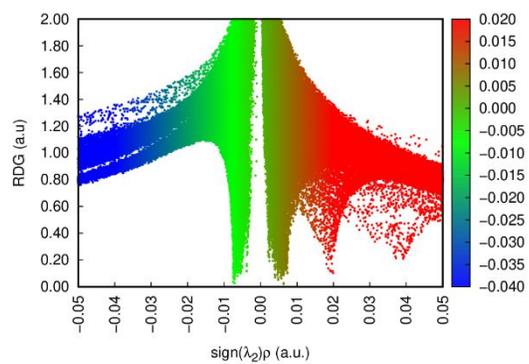


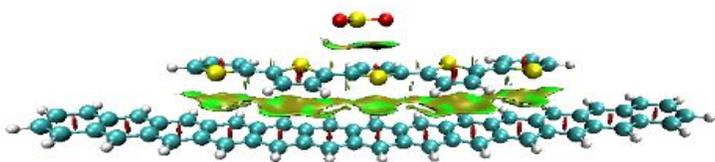
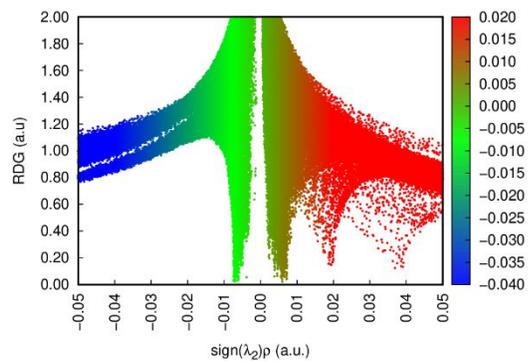
Figure S12: DOS spectra of (a) $C_{54}H_{30} \dots 9PT \dots CO$, (b) $C_{54}H_{30} \dots 9PT \dots NH_3$, (c) $C_{54}H_{30} \dots 9PT \dots SO_2$ and (d) $C_{54}H_{30} \dots 9PT \dots NO_2$ composite-analyte complexes



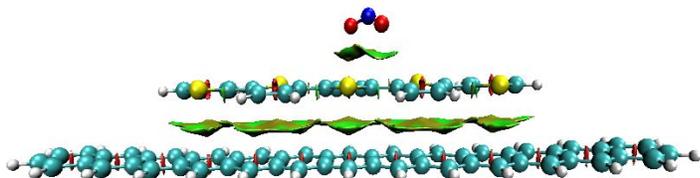
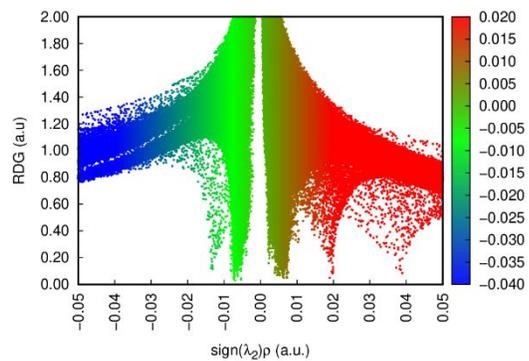
(a)



(b)



(c)



(d)

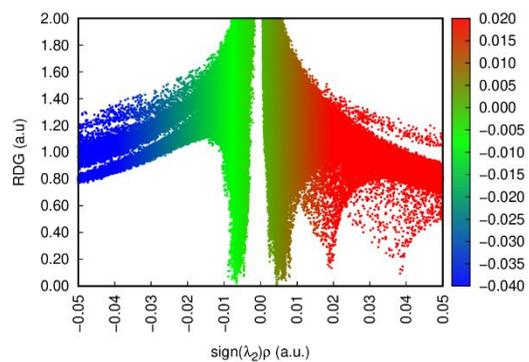
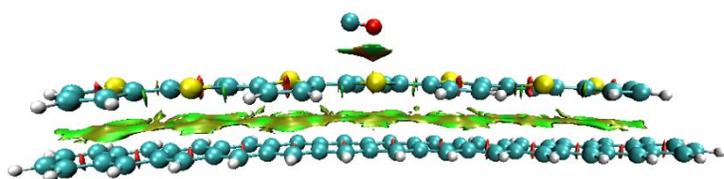
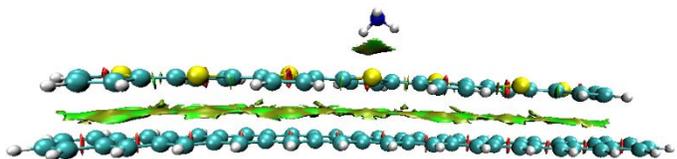
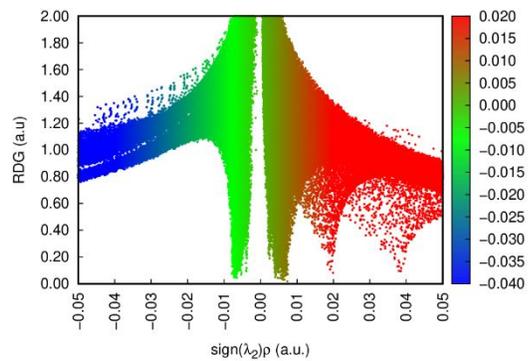


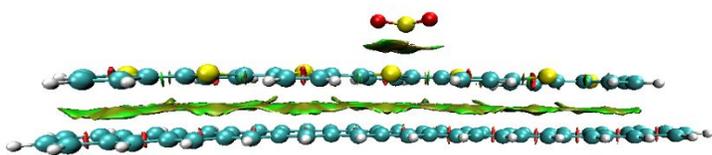
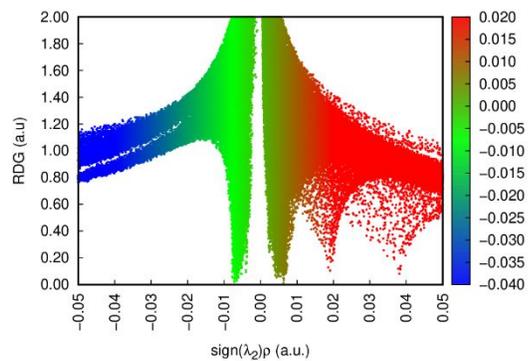
Figure S13: Color-mapped RDG isosurface graphs (3D) and scatter diagrams (2D) of (a) $C_{54}H_{30} \dots 5PT \dots CO$, (b) $C_{54}H_{30} \dots 5PT \dots NH_3$, (c) $C_{54}H_{30} \dots 5PT \dots SO_2$ and (d) $C_{54}H_{30} \dots 5PT \dots NO_2$ composite-analyte complexes



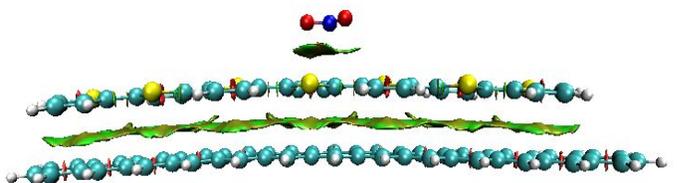
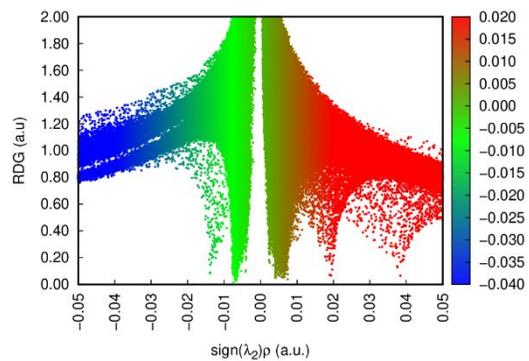
(a)



(b)



(c)



(d)

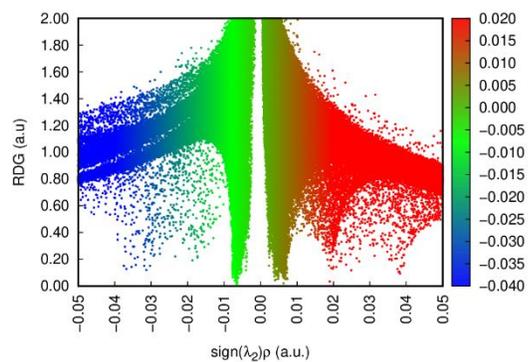


Figure S14: Color-mapped RDG isosurface graphs (3D) and scatter diagrams (2D) of (a) C₅₄H₃₀ ... 7PT ... CO, (b) C₅₄H₃₀ ... 7PT ... NH₃, (c) C₅₄H₃₀ ... 7PT ... SO₂ and (d) C₅₄H₃₀ ... 7PT ... NO₂ composite-analyte complexes

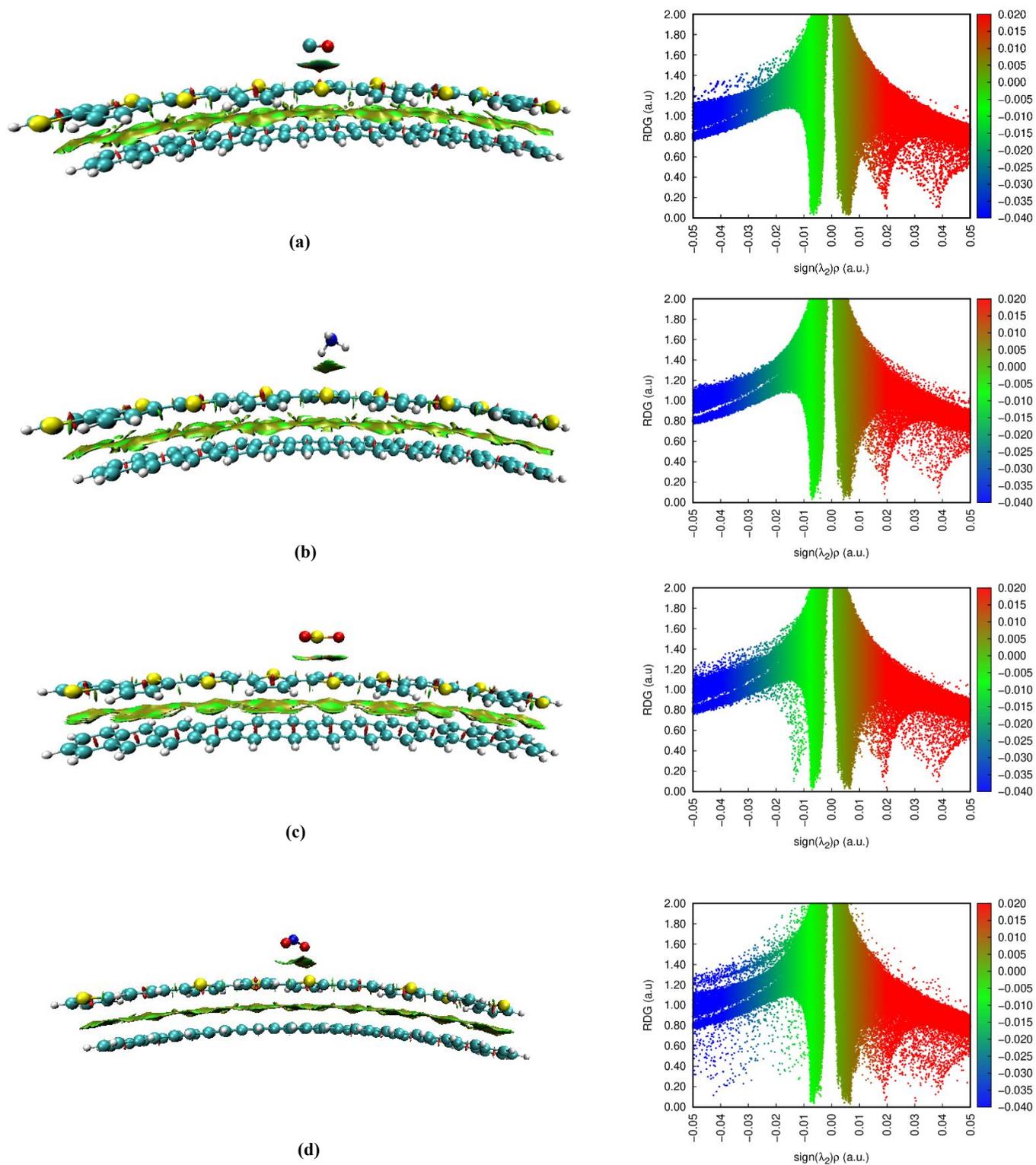


Figure S15: Color-mapped RDG isosurface graphs (3D) and scatter diagrams (2D) of (a) $C_{54}H_{30} \dots 9PT \dots CO$, (b) $C_{54}H_{30} \dots 9PT \dots NH_3$, (c) $C_{54}H_{30} \dots 9PT \dots SO_2$ and (d) $C_{54}H_{30} \dots 9PT \dots NO_2$ composite-analyte complexes