

BOUND RO-VIBRATIONAL STATES OF H₂...CN(X²Σ⁺) VAN DER WAALS COMPLEX

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The abstraction reaction H₂+CN → H+HCN proceeds via a collinear transition state. The entrance channel to this transition state may be examined through spectroscopic studies of the H₂-CN van der Waals complex. In addition, as the barrier to reaction is only 1200 cm⁻¹, it may be possible to initiate reaction within the cluster by vibrational excitation of the H₂ moiety. To learn more about the pre-reaction dynamics and identify states that sample the transition state geometry, we have examined the characteristics of bound states supported by the van der Waals well.

A previously reported 4-D interaction potential (with H₂ and CN bonds fixed) was used to calculate the bound states for **J**=0,1,..., ignoring spin. The ro-vibrational eigenstates are calculated in a body-fixed formalism, where the unsigned projection of **J** onto van der Waals bond (*K*) and its reflectional parity (*ε*) are nearly good quantum numbers. For the *para*-H₂ complex the lowest energy state is *K*=0⁺ corresponding to the **J**=0 manifold. Its binding energy with respect to the H₂(*j*=0)+CN(*j*=0) asymptote is ~16 cm⁻¹. Similarly, the *ortho*-H₂ complex has a *K*=0⁺ ground state deriving from **J**=0. It is bound by ~31 cm⁻¹ relative to the H₂(*j*=1)+CN(*j*=0) asymptote. In both cases, the first excited state is only ~1 cm⁻¹ above the zero point; it derives from **J**=1 and belongs to *K*=0⁻ symmetry with some mixing from *K*=1⁻ state. Potential and Coriolis coupling terms mix different *K* and *ε* states, rendering the eigenstate structure very complicated. Examination of probability density for the *ortho*-H₂ complex showed that some low-lying states sample the linear H-H...C-N geometry.