

MILLIMETER WAVE SPECTROSCOPY OF THE VINYL-D (H_2CCD) RADICAL GENERATED BY UV LASER PHOTOLYSIS IN A PULSED SUPERSONIC JET EXPANSION. DETERMINATION OF THE PROTON TUNNELING SPLITTING

MASATO HAYASHI, KENSUKE HARADA, KEIICHI TANAKA, and TAKEHIKO TANAKA, *Department of Chemistry, Faculty of Science, Kyushu University 33, Hakozaki, Higashiku, Fukuoka 812-8581, Japan;* RICHARD LAVRICH, *National Institute of Standards and Technology, USA.*

The rotational and proton tunneling spectra of the vinyl-d (H_2CCD) radical have been observed by millimeter wave spectroscopy combined with a pulsed supersonic jet technique. H_2CCD was generated by the 193 nm excimer laser photolysis of vinyl chloride- d (H_2CCDCl). The pure rotational transitions (*a*-type transitions), $N_{K_a K_c} = 1_{01} - 0_{00}, 2_{02} - 1_{01}, 3_{03} - 2_{02}, 2_{11} - 1_{10}, 2_{12} - 1_{11}$ in the 0^+ and 0^- states, were observed in the frequency region of 52.9–158.7 GHz. The $0^+ \leftarrow 0^-$ and $0^- \leftarrow 0^+$ proton tunneling transitions (*b*-type transitions), $N_{K_a K_c} = 1_{11} - 0_{00}, 1_{10} - 1_{01}, 2_{11} - 2_{02}, 2_{12} - 2_{01}$, were observed in the frequency region of 184.7–236.8 GHz. The observed rotational lines were split into fine and hyperfine components due to the spin-rotation interaction and the spin- nuclear spin interaction of the acetylenic (CD) deuteron (α -deuteron) and the methylenic (CH_2) protons (β - protons). The molecular constants such as rotational constants, spin- rotation interaction constants, and hyperfine interection constants, were determined by a least squares fitting of the observed spectrum, together with the proton tunneling splitting $\Delta E_0 = 1164.861(20)$ MHz. The barrier height of the double minimum potential was estimated to be 1520 cm^{-1} from the observed tunneling splitting assuming a one dimensional model. This potential barrier height is consistent with that of the normal species (H_2CCH), 1580 cm^{-1} , derived from the tunneling splitting reported to be $16271.8429(59)$ MHz ^a, if the zero point energy is taken into consideration.

^aK. Tanaka, M. Toshimitsu, K. Harada, T. Tanaka, *J. Chem. Phys.* **120**, 3604 (2004)