

ROTATIONAL SPECTRA AND STRUCTURE OF H₂S COMPLEXES: Ar(H₂S)₂ AND C₂H₄-H₂S

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Recently, we determined hydrogen bond radii for HF, HCl, HBr, HI, HCN, H₂O and HCCH, using the accurate structural data available from microwave spectroscopic studies on several B:HX complexes, where B is a hydrogen bond acceptor^a. For H₂S, the available data is limited compared to other hydrogen bond donors listed above. Hence, a systematic investigation has been started on several H₂S complexes. Several transitions have been observed for Ar-(H₂S)₂ and C₂H₄-H₂S complexes. The rotational spectrum of C₂H₄-H₂S complex appears to be a composite of rotational spectra of the hydrogen bonded C₂H₄-H₂O and the van der Waals complex C₂H₄-Ar. Each transition is split in to four with a smaller splitting of about 0.14 MHz and a larger splitting of 1.67 MHz in (B+C)/2. The rotational constants for the strongest progression are A = 26(1) GHz; B = 1972.88(1) MHz and c = 1866.68(1) MHz. Assignment of Ar-(H₂S)₂ spectra is in progress. It was realized that the rotational spectrum for (H₂S)₂ complex has not been completely solved yet^b and hence a 'revisit' started. Several more progressions have been assigned for this dimer now. The search for deuterated isotopomers was carried out by bubbling H₂S through D₂O and some new H₂O-H₂S transitions have been observed. During this talk, rotational spectra of all these complexes will be discussed.

^aP. K. Mandal and E. Arunan J. Chem. Phys. 114, 3880 (2001)

^bF. J. Lovas private communication