

HIGH RESOLUTION LASER EXCITATION SPECTROSCOPY OF SrCH₃, CaBH₄ AND SrBH₄

M. J. DICK, *Department of Physics, University of Waterloo, 200 University Ave. West, Waterloo, ON, N2L 3G1 Canada*; P. M. SHERIDAN, J.-G. WANG and P. F. BERNATH, *Department of Chemistry, University of Waterloo, 200 University Ave. West, Waterloo, ON, N2L 3G1 Canada*.

High resolution laser excitation spectroscopy has been used to record the $\tilde{A}^2E - \tilde{X}^2A_1$ electronic transition of SrCH₃ in a laser ablation/molecular jet source. SrCH₃ was synthesized by the reaction of UV-ablated strontium atoms with a 1% mixture of tetramethyl tin in argon. Transitions arising from the $K'=1 \leftarrow K''=0$, $K'=0 \leftarrow K''=1$ and $K'=2 \leftarrow K''=1$ sub-bands have been observed and assigned. Rotational and fine structure parameters have been determined for the \tilde{A}^2E state. An analysis of the spin-orbit and spin-rotation constants indicates that the \tilde{A}^2E state does not arise entirely from an atomic orbital of p character. In addition, the Jahn-Teller coupling was found to be negligible in the zero-point vibrational level of the \tilde{A}^2E state. The rotational constants were used to estimate a structure for SrCH₃ in the \tilde{A}^2E state. The geometry changes observed between the \tilde{X}^2A_1 and \tilde{A}^2E states of SrCH₃ are similar to those observed for CaCH₃. In addition, the $\tilde{B}^2E - \tilde{X}^2A_1$ transitions of CaBH₄ and SrBH₄ have been observed. The metal borohydrides were synthesized in a jet source by reaction of UV-ablated metal atoms with a 5% mixture of diborane in argon. A rotational analysis is currently in progress and a comparison of the spectroscopic and structural parameters for both molecules will be presented.