

MEASUREMENT OF THE ELECTRON BINDING ENERGY OF SIZE SELECTED $\text{IBr}^-(\text{CO}_2)_n$ CLUSTERS THROUGH ANION PHOTOELECTRON IMAGING

RYAN M. D. CALVI, DJANGO H. ANDREWS, and W. CARL LINEBERGER, *JILA and Department of Chemistry and Biochemistry, University of Colorado and National Institute of Standards and Technology, Campus Box 440, University of Colorado, Boulder, Colorado 80309-0440.*

We report the measurement of the electron binding energies of size selected $\text{IBr}^-(\text{CO}_2)_n$, $n < 10$ cluster anions. The data are interpreted using *ab initio* calculations and Franck-Condon simulations to assist in the assignment of the photoelectron spectra. The $\text{IBr}(\text{CO}_2)_n$ anion are formed by a 1 kV electron impact on dilute IBr and CO_2 in a pulsed supersonic expansion. Time-of-flight mass spectrometry allows a size-selected cluster anion to interact with a pulsed ultraviolet laser radiation tunable between 210 and 320 nm. The photodetached electrons are energy analyzed using a photoelectron imaging spectrometer operating in the velocity map imaging mode. The spectrometer determines electron kinetic energy, with an energy resolution of 2.5 percent. Trends in electron binding energy as a function of size will be discussed in regard to the trend that develops upon the addition of each cluster, and compared to previous photoelectron spectroscopy of $\text{I}_2(\text{CO}_2)_n$ clusters. Supported by NSF and AFOSR.