## THE <sup>1</sup>II STATES OF NaCs: SPECTROSCOPY, LIFETIMES, PERMANENT AND TRANSITION DIPOLE MOMENTS.

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The NaCs molecule is one of the prospective objects for production of ultracold polar molecules. LIF Fourier transform spectroscopy study is presented for the  $(1,3)^1\Pi$  states with 0.03 cm<sup>-1</sup> resolution. Potential energy curves (PECs) are obtained by the Inverted Perturbation Approach reproducing  $(3)^1\Pi$  state<sup>*a*</sup> energies for R = 3 to 11 Å. In the B(1)<sup>1</sup> \Pi state we accounted for numerous B(1)<sup>1</sup>  $\Pi \sim c^3 \Sigma^+$  perturbations by omitting perturbed levels for the fit to construct the PEC for R = 2.6 to 8.4 Å.

The permanent electric dipole moments d and the  $\Lambda$ -splitting were measured<sup>b</sup> by dc Stark mixing and electric RF-optical double resonance methods yielding d within 5 - 8 D for  $(3)^1 \Pi$  and  $d \sim 1$  D for the D(2)<sup>1</sup>  $\Pi$  ( $\nu < 3$ ) state. The radiative lifetimes  $\tau$  were measured from LIF kinetics as  $\tau = 29$  to 21 ns for  $(3)^1 \Pi$  ( $\nu = 3$  to 25) and  $\tau = 37$  ns for D(2)<sup>1</sup>  $\Pi$  ( $\nu = 0$ ).

The measured data are supported by electronic structure calculations for the  $(1-3)^1\Pi$  states<sup>b</sup> by many-body multipartitioning perturbation theory of PECs, permanent and transition dipole moments, as well as angular coupling matrix elements for the lowest singlet states. The predicted *d* values reproduce their experimental counterparts within the measurement errors. Lifetimes for the  $(1-3)^1\Pi$ -states have been calculated in Hund's "a" coupling case using the approximate sum rule over the lower vibronic states. The spectra and formation rates of ultracold NaCs in the  $X^1\Sigma^+$  (v = 0, J = 0) state were simulated for the optical cycle  $a^3\Sigma^+ \rightarrow B(1)^1\Pi \sim c^3\Sigma^+ \sim b^3\Pi \rightarrow X^1\Sigma^+$ . The Riga team and the Moscow team acknowledge support by NATO SfP 978029 Optical Field Mapping grant, the Hannover team support by the DFG through the SFB 407.

<sup>&</sup>lt;sup>a</sup>O. Docenko et al J. Chem. Phys., <u>124</u>, 174310, 2006.

<sup>&</sup>lt;sup>b</sup>J. Zaharova et al J. Chem. Phys., <u>124</u>, 184318, 2006.