KINETICS MEASUREMENTS OF HO_2 AND RO_2 SELF AND CROSS REACTIONS USING INFRARED KINETIC SPECTROSCOPY (IRKS)

A. C. NOELL, L. S. ALCONCEL, D. J. ROBICHAUD, and M. OKUMURA, *Department of Chemistry, California Institute of Technology, Pasadena, CA 91125*; S. P. SANDER, *Jet Propulsion Laboratory, Pasadena, CA 91109.*

Reactions between RO_2 and HO_2 are important in the gas phase oxidation of volatile organics in the atmosphere. The rate constants for the reactions of HO_2 + $C_2H_5O_2$, and $C_2H_5O_2$ + $C_2H_5O_2$ were measured using a novel approach to radical radical reactions that employs two independent spectroscopic probes. HO_2 was monitored at 6638.2 cm^{-1} using near-IR diode laser WM spectroscopy, and $C_2H_5O_2$ was monitored at 250 nm using UV absorption. Experiments were run under conditions relevant to the upper troposphere (221-298 K and 50-400 Torr). Measurements of the rate constants for both reactions as well as a measurement of the branching fraction leading to the alkoxy channel of the $C_2H_5O_2$ self reaction will be discussed. Preliminary results are: $k_{HC} = (6.4 \pm 1.4) \times 10^{-13}$ cm^3 $molec^{-1}$ s^{-1} Exp((601 ± 120)/ T), $k_{CC} = (1.10 \pm 0.20) \times 10^{-13}$ cm^3 $molec^{-1}$ s^{-1} independent of temperature, and a branching fraction of 0.32 ± 0.13.