The photophysics of pyrimidine bases following excitation at 267 nm were studied using femtosecond IR and UV/Vis spectroscopy. A majority of excited molecules rapidly undergo internal conversion to the electronic ground state, which is formed with \( \approx 37\,000\,\text{cm}^{-1} \) of excess vibrational energy. Vibrational cooling of this hot ground state was followed via recovery of the thermalized ground state spectrum. Vibrational cooling times obtained from the recovery of electronic vs. vibrational bands differ and are sensitive to the isotopic composition of the solvent. These results suggest that intramolecular vibrational redistribution is 10-100 times slower than expected in these molecules. A fraction of molecules in the initially excited \( \pi\pi^* \) state decay to other electronic excited states. Vibrational bands of these states are observed at \( \approx 1600\,\text{cm}^{-1} \) and \( \approx 1750\,\text{cm}^{-1} \) and are assigned to triplet and singlet \( n\pi^* \) states, respectively. While the electronic absorption bands of these states have been observed previously, the greater resolution provided by their vibrational spectra provides new insights into excited state photophysics of pyrimidine bases.