

INFRARED SPECTROSCOPY OF THE INTERACTION OF O₂ WITH NO TRAPPED IN SOLID NEON

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When O₂ trapped in solid neon at 4.3 K in sites adjacent to an NO molecule is excited by 240 to 420 nm radiation, NO₃ is stabilized. Photoexcitation of NO₃ to its 663-nm state, which has ²E' symmetry, leads to the re-formation of weakly interacting (O₂)(NO) pairs trapped in solid neon. In addition to infrared absorptions near the gas-phase band centers of NO and O₂, the weakly bound complex shows several absorptions at lower frequencies. Experiments in which either the O₂ or the NO moiety of the initial sample deposit is enriched in O-18 show O-isotopic randomization of the complex which results from photodestruction of the NO₃ product.