

## LASER-INDUCED FLUORESCENCE SPECTROSCOPY OF THE $\tilde{A}^2\Pi - \tilde{X}^2\Sigma^+$ TRANSITION OF SrCCH AND SrCCD

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The family of molecules consisting of an alkaline earth metal atom bonded to an organic ligand has been under intensive investigation recently, since these molecules play a significant role in synthesis of organic compounds, and since some members like MgNC and MgCN have been detected in stellar atmospheres. Our group has previously studied CaCCH, MgCCH, CaOCH<sub>3</sub>, and SrOCH<sub>3</sub>, and in this work we consider the  $\tilde{A}^2\Pi - \tilde{X}^2\Sigma^+$  transition of SrCCH and SrCCD. The molecules were created in a pulsed-jet laser ablation source through reaction of strontium atoms ablated from a solid rod with methane gas entrained in helium. High-resolution spectra were collected through laser-induced fluorescence following excitation of the molecules formed in the jet with a cw ring dye laser. Results of the analysis are compared to those of a previous high-resolution study of SrCCH.<sup>a</sup> The data provide information on the C-H bond length via the isotopic substitution, and the lambda-doubling parameter  $p$  allows for an estimate of the position of the  $\tilde{B}^2\Sigma^+$  state for both isotopes.

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<sup>a</sup>M. J. Dick, P. M. Sheridan, J.-G. Wang, and P. F. Bernath, *J. Mol. Spectrosc.* **233**, pp. 197-202 (2005)