

MILLIMETER-WAVE SPECTROSCOPY OF THE D₂CCD RADICAL

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The tunneling-rotation transitions of the vinyl-*d*₃(D₂CCD) radical produced by the 193 nm excimer laser photolysis of vinyl-*d*₃ chloride(D₂CCDCl) have been observed by millimeter-wave spectroscopy combined with a pulsed supersonic jet technique. The *b*-type transitions, $N_{K_a K_c} = 1_{11} - 0_{00}$, $1_{10} - 1_{01}$, $2_{12} - 1_{01}$, for both the $0^- \leftarrow 0^+$ and $0^+ \leftarrow 0^-$ tunneling subbands were observed in the frequency region of 101-183 GHz, split into fine and hyperfine components due to the spin-rotation interaction and the spin-spin interaction of the α (CD)- and the β (CD₂)-deuterons. The molecular constants such as rotational constants, spin-rotation interaction constants, and hyperfine interaction constants were determined by a simultaneous analysis of the observed spectrum and previously reported $J=1 \leftarrow 0$ pure rotational transition^a together with the tunneling splitting $\Delta E_0 = 771.843(23)$ MHz between the 0^+ and 0^- states. The tunneling splitting for D₂CCD is less than 1/20 of that for H₂CCH (16 271.842 9(59) MHz)^b due to the mass effect of the α -H and D, and it is about 2/3 of H₂CCD (1 186.820(26) MHz) indicating the mixing of the vibrational modes for α -H/D and β -H₂/D₂. From the observed tunneling splitting, the barrier height h of the double minimum potential for D₂CCD was estimated to be 1549 cm⁻¹ using one dimensional model. The barrier height h for D₂CCD is almost the same as those for H₂CCH and H₂CCD, 1580 and 1520 cm⁻¹, respectively, as expected by B.O. approximation and the isotopic effect due to zero point energies. The off-diagonal Fermi interaction constant, δa_F , which is responsible to the mixing of the wavefunctions of *ortho*($I_\beta = 0, 2$) and *para*($I_\beta = 1$) states, has been determined to be 19.8(30) MHz. The off-diagonal Fermi interaction may cause the nuclear spin conversion between the *ortho*- and *para*-states for D₂CCD.

^aE. Kim, S. Yamamoto, *J. Chem. Phys.* **116**, 10713, (2002).

^bK. Tanaka, M. Toshimitsu, K. Harada, T. Tanaka, *J. Chem. Phys.* **120**, 3604, (2004).