

LARGE AMPLITUDE MOTIONS AND INFORMATION TRANSFER ALONG CONJUGATED BONDS: THE CASE OF PARATOLUALDEHYDE

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Historically, barriers to large amplitude motions are often related, using steric hindrance arguments, to the local atom arrangement. More recently, large changes have been found in the barrier inhibiting methyl rotation, when electronic excitation or ionization of a substituted toluene occurs,^a suggesting molecular orbital occupancy as a dominant factor in aromatic molecules. Theoretically, the relation can also be made to the molecular orbital and electron density structure.

Experimentally, we examine the situation by comparing toluene with paratolualdehyde. In particular, the barrier in toluene, $CH_3 - C_6H_5$, is six-fold by symmetry. In paratolualdehyde, $CH_3 - C_6H_4 - CHO$, the aldehyde group is far enough from the methyl rotor that direct through-space interactions should be rather small. Thus, a three-fold contribution to the barrier would leave electron orbital effects in the π -system as the primary causal agent.

Microwave spectroscopy is well suited for discriminating between V_3 and V_6 contributions - but only if torsionally excited states can be accessed which typically requires temperatures much higher than those encountered in pulsed-jet expansions. Thus, the supersonic-jet FT-MW spectrometers in Hannover and Gaithersburg as well as the free-jet CW mm-wave spectrometer in Bologna were needed to take the spectra. Analyzing 5 torsional species using the Belgi program, the barrier was found to vastly be dominated by the V_3 vs. a minor V_6 contribution, thus revealing the information transfer through conjugated π -electron systems.

^aL. H. Spangler, *J. Rev. Phys. Chem.* 48 (1997) 481-510