The Higher K_{-1} States of Hydrogen Peroxide*

We previously reported (1) the spectrum of hydrogen peroxide between 80 and 700 GHz that originates from transitions between the $K_{-1}=0$, 1, and 2 levels of $\tau=1$, 2 and the K=0 and 1 levels of $\tau=3$, 4. We have also discussed the analysis of this spectrum to microwave accuracy (\sim 0.1 MHz) by means of a centrifugal distortion Hamiltonian in which the only effect of the molecular internal rotation is the \sim 11-cm⁻¹ splitting between the $\tau=1$, 2 and $\tau=3$, 4 levels. Because the data set was limited to low K_{-1} , it was not possible to calculate Δ_K , and in our analysis Δ_K was fixed to an earlier value derived from infrared studies. Since Δ_K is highly correlated with the A rotational constant and W, the torsional splitting, inaccuracies in Δ_K severely affect the values of the other two constants and make predicted line frequencies at higher values of K_{-1} much less accurate.

Higher- K_{-1} transitions are observable in the 80- to 700-GHz region, but because they are high J, relatively weak, and mixed with a dense background of lines that originate in low-lying excited torsional states, they are difficult to assign. However, we have assigned and analyzed an excellent fir Fourier transform spectrum of hydrogen peroxide provided to us by Dr. Kelley Chance of the Harvard-Smithsonian Astrophysical Observatory. The results of this analysis, combined with our earlier millimeter and submillimeter work, have made it possible to measure extensive new higher- K_{-1} data in the millimeter and submillimeter spectral range. We have now measured 101 new lines that, added to our original data set, give a total of 284 new lines in the 80- to 700-GHz region and include $K_{-1} = 0$, 1, 2, 3, 4 of $\tau = 1$, 2 and K = 0, 1, 2, 3 of $\tau = 3$, 4. We have performed an analysis similar to that of our earlier work

We previously described both the experimental and theoretical techniques used for our analysis of hydrogen peroxide, and interested readers are referred to Ref. (1) for details. Table I shows our new data, and Fig. 1 displays the total data set in the form of a FORTRAT diagram. It can be seen that our new data are $\Delta J = \pm 1$ and that the branches originate in the terahertz region.

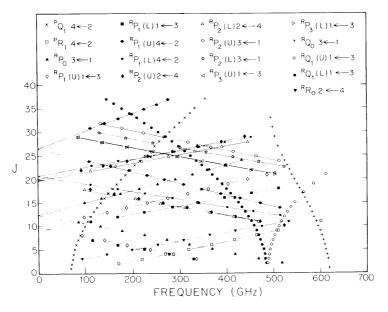


FIG. 1. FORTRAT diagram of the observed branches of HOOH.

^{*} This work supported by NASA Grant NSG-7540.

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TABLE I

New Observed Transitions of HOOH (MHz)

R _{F2} (Upper) Branch Transitions		R _{F2} (Lower) Branch Transitions		
Transition	Frequency	Transition	Frequenc	
r + 2 - r + r		<u>τ = 2</u>	Observed	
9 ₃₇ - 10 ₂₁ 10 ₂₀ - 11 ₀	524 038.11 472 527.24	1037 - 1129	470 905. 3	
	0 472 327.24 , 421 037.05	11 ₃₈ - 12 ₂₁₀	416 770.2	
	1 369 576, 78	12 ₃₉ + 13 ₂₁₁	3dd 491.7	
12310 - 132	2 308 370, 78	13 ₃₁₀ - 14 ₂₁₂	314 052.0	
3311 - 142	3 318 156, 12	14 ₃₁₁ - 15 ₂₁₃	231 432.5	
*312 * 10 ₂	4 266 785, 43	15 ₃₁₂ + 16 ₂₁₄	208 613.0	
5 ₃₁₃ - 16 ₂	5 215 475.52	16 ₃₁₃ - 17 ₂₁₅	155 574, 8	
.6314 - 172	6 164 237, 48	1/314 + 18 ₂₁₃	102 29a, b	
$r_{315} \leftarrow 18_{9}$	7 113 082.96	18 ₉₁₅ 19 ₉₁₇	(48.7.4.8	
.8 ₃₁₆ ← 19 ₉ .	A (52 024.01)	19 ₃₁₅ → 20 ₂₁₈	(5 068.3	
.9 ₉₁₇ - 20 ₉ .	9 (11 072, 81)	20317 - 21919	(.9 194. 8	
U ₃₁₈ - 21 ₂	n (39.757,85)	$21_{318} \rightarrow 22_{220}$	113 643, 8	
$1_{319} - 22_{9}$	1 90 455, 10	22 ₉₁₀ - 23 ₉₉₁	168 434.7	
Z320 → Z33	2 141 005, 86	23 ₃₂₀ → 24 ₂₂₂	223 584.8	
$3_{321} \rightarrow 2_{42}$	3 191 396, 73	24 ₃₂₁ - 2: 223	279 109.8	
4322 - 252	4 241 614, 15	25399 - 26994	330 023.2	
$5_{323} - 20_{23}$	5 291 644, 55	26 ₃₂₃ → 27 _{22:}	391 335,90	
5994 - 2799	a 341 474.11	27 ₃₂₄ → 28 ₂₂₅	448 056.0	
$7_{325} + 28_{25}$	7 391 089.33	000		
B ₃₂₆ → 29 ₂₅	8 440 476.17			
9327 - 302	489 620.96			
		R _L (Lower) Bran	on Transitions	
R _{F1} (Upper) F	ranch Transitions	R _F (Lower) Bran	1. 0.1510118	
Transition	Frequency	Transition	Frequency	
:4 +=	Observed	т = 4 т = 2	Observed	
2211 - 131	3 555 548, 40	1129 - 12111	51 6 987. d	
$^{3}212 - ^{14}1$	4 512 539, 44	12 ₂₁₀ - 13 ₁₁₂	460 020.8	
$4_{213} \leftarrow 15_1$	470 178, 32	13 ₉₁₁ + 14 ₁₁₉	402 798.93	
$5_{214} \leftarrow -16_{1}$.: 428 474.90	14919 - 15114	34:- 361.1	
$s_{210}^{219} - 17_{11}^{11}$	3 87 438, 99	1: 213 - 16115	287 749.53	
7 ₂₁₆ - 18 ₁	347 080.11	18 ₂₁₄ + 17 ₁₁₆	230 008. 34	
1210 - 111 8- + 19	307 408, 36	17 ₂₁₅ + 18 ₁₁₇	172 184. 4	
8 ₂₁₇ + 19 ₁ 9 + 20	268 434.20	18 19	114 326, 93	
9 ₂₁₈ + 20 ₁₁	0 230 167, 83	18 ₂₁₀ - 19 ₁₁₈	(06 486, 8	
0219 - 21	1 192 619, 92	19 ₂₁₇ - 20 ₁₁₉	(1 282. 6	
1000 - 22,4	2 192 519, 92	20 ₂₁₈ → 21 ₁₂₀	(1.282.5	
2221 + 2315	3 100 801.20	21 ₂₁₉ - 22 ₁₂₁		
13 ₂₂₂ + 24 ₁₅	4 119 722, 49	22 ₂₂₀ - 23 ₁₂₂	116 389, 49	
⁴ 223 - ²⁵ 13	5 (84.394,50)	23001 - 24	173 613. 4	
$5_{224} \leftarrow 26_{13}$	2 (49 828.12)	$24_{999} \rightarrow 25_{194}$	230 540, 31	
5 ₉₉₅ - 27 ₁₆	7 (16 034.22)	25 ₀₀₀ → 26 ₁₀	287 111. 4	
7 ₂₂₄ → 28 ₁₅	8 (16.976.47)	$26_{224} - 27_{126}$	343 267.3	
8 ₂₂₇ → 29 ₁₅	(49 193, 19)	27 ₂₂₅ → 28 ₁₂₇	398 948, 70	
9 ₂₂₈ → 30 ₁₅	n (80 605.36)	28 ₂₂₆ - 29 ₁₂₈	454 096, 5	
U ₉₉₀ → 31 ₁ ,	, 111 202.63			
1 ₉₃₀ - 32,	2 140 974.77			
$^{12}231 \rightarrow ^{33}13$	3 169 911, 59			
3232 - 3415	4 198 003, 20			
4 ₂₃₃ - 30 ₁₅	h 225 240, 10			
5 ₂₃₄ → 35 ₁ ;	d 251 612.75			
5 ₂₃₅ - 37 ₁₃	7 277 112.13			
		n () - 1 =	in many to	
κ _{P2} (Upper) Ε	ranch Transitions	R _{P3} (Upper) Brand	an iransilions	
Transition	Frequency	Transition	Frequency	
- 3 т =	1 Observed	т = 1 т = 3	Observed	
3 ₃₂₁ + 24 ₂		20417 - 21319	505 478.0	
³²¹ - ²⁵ 2	461 893.54	21 ₄₁₈ - 22 ₃₂₀	453 390. 2	
24 ₃₂₂ - 25 ₂	413 861, 31	22 ₄₁₉ + 23 ₃₂₁	401 269.5	
26 ₃₂₃ + 26 ₂ 36 ₃₂₄ + 27 ₂	366 148.00	$22_{419} \leftarrow 23_{321}$ $23_{420} \leftarrow 24_{322}$	349 116.6	
35324 - 27 ₂			296 932, 3	
27 ₃₂₅ - 28 ₂	271 755.29	24 ₄₂₁ + 25 ₃₂₃	244 717.5	
18 ₃₂₆ + 29 ₂	28 271 755, 29 225 115, 02	25 ₄₂₂ + 26 ₃₂₄	192 473, 4	
19 ₃₂₇ - 30 ₂	190 000 10	26 ₄₂₃ - 27 ₃₂₅	140 201.5	
SU ₉₀₆ - 31 ₉	178 872.17	27 ₄₂₄ + 28 ₃₂₆	87 903. 3	
329 + 32 ₂	31 133 047.01	28 ₄₂₅ - 29 ₃₂₇	87 903, 3	
R _{P2} (Lower) I	branch Transitions	R _{P3} (Lower) Bran	cn Transitions	
Transition	Frequency	Transition	Frequenc	
· - 3 r -		т = 1 т = 3	Observed	
2319 - 232		20416 - 21318	505 076, 4	
23 ₉₉₀ - 24 ₉	11 468 587, 22	21 ₄₁₇ - 22 ₃₁₉	452 862.1	
	22 413 361, 72	219 ± 23	400 :83.8	
14 ₃₂₁ - 25 ₂	357 684, 42	22 ₄₁₈ + 23 ₃₂₀	348 236. 1	
	24 307 984, 42	23 ₄₁₉ + 24 ₃₂₁		
10q93 - 4/9	301 540.89	24 ₄₂₀ - 25 ₃₂₂	295 813. 2	
$7_{324} - 26_2$	244 920, 57	25 ₄₂₁ + 26 ₃₂₃	243 308.9	
:8 + 29_	₂₇ 187 816.91	$26_{422} - 27_{324}$	190 715, 8	
8 ₃₂₅ + 29 ₂ 9 ₃₂₆ - 30 ₂	130 227. 97	27 ₄₂₃ - 28 ₃₂₅	138 026. 2	

^a Numbers in parentnesis are calculated from the constants of Table II.

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TABLE II
Rotational Constants (MHz)

	т = 1, 2		τ = 3, 4		
	value	σ	value	σ	
А	301 874.205	0.037	301 586.074	0.075	
В	26 212.439	0.046	26 155.639	0.036	
С	25 098.604	0.046	25 186. 461	0.036	
Δ _J (· 10 ⁰)	0.105292	0.0000029	0.0996169	0.000014	
^ (·10 ⁻¹)	0.112180	0.000026	0.115275	0.000051	
$\Delta_{K} \stackrel{\text{10}}{(\cdot 10^{3})}$ $\delta_{J} \stackrel{\text{10}}{(\cdot 10^{3})}$	0.120340	0.00012	0.118558	0.00023	
$\delta_{1}^{1}(\cdot 10^{3})$	-0.258456	0.0014	0.686780	0.0012	
δ _K (·10 1)	0.978631	0.0027	0.6 38 326	0.00021	
H _{IK} (· 10 ³)	0.506307	0.0097	0.0594176	0.0013	
H _w (· 10 ²)	-0.161735	0.0087	- 0. 0506399	0.0073	
H _K (⋅10 ²)	0.185268	0.097			
h _J (· 10 ⁷)	0.304918	0.022	-0.515225	0.011	
n _{JK} (·10 ³)	0.172789	0.00090	0.129727	0.0011	
h _K (· 10 ¹)	0.440464	0.038	-0.631712	0.0028	
L _{JJK} (-10 ⁷)	-0.159961	0.0045	- 0.105713	0.0071	
L _{IK} (·10 ⁵)	- 0.933110	0.051	0.829787	0.0077	
L _{KKJ} (· 10 ⁴)	0.218515	0.038			
rms	0.126		0.112		
number of data points	119		141		
W	342 881.62				

The analysis of hydrogen peroxide is complex because of the effects of the torsional motion. This motion splits the ground vibrational state into two pairs of sublevels, $\tau=1,2$ and $\tau=3,4$, and gives the selection rules, $\tau=1\leftrightarrow 3$ and $\tau=2\leftrightarrow 4$, in addition to the usual asymmetric rotor restrictions. As before, we analyze the lower $\tau=1,2$ levels by means of combination differences connected in the $\tau=3,4$ state. We assume that the rotational constants in $\tau=1$ and $\tau=2$ are the same and find that the 119 distinct combination differences can be analyzed with an rms deviation of 0.13 MHz. The values of the constants that result from this analysis are shown in Table II. Fewer combination differences exist in $\tau=3,4$, so we supplement these data. Additional energy levels in $\tau=3,4$ are calculated from combination of the energy levels produced by the $\tau=1,2$ analysis with measured transition frequencies between $\tau=1,2$ and $\tau=3,4$. We are careful to use only $\tau=1,2$ states that we believe to be well established. The effect of this procedure is that some lines at high J (>30) are not used in the analysis because they were not part of the $\tau=1,2$ analysis and that these transitions may not be reproduced to microwave accuracy. While it would be possible to include them in the analysis, we feel that this is undesirable because the $\tau=3,4$ analysis would have to readjust its constants to cancel extrapolation errors in the $\tau=1,2$ analysis. The results of the $\tau=3,4$ analysis are also shown in Table II.

With our new measurements the characterization of the rotational-torsional spectrum of the $\tau=1,2,3,4$ states of hydrogen peroxide in the spectral region below 700 GHz is virtually complete. Most transitions between thermally populated energy levels have been directly measured and those that have not (these are mostly low-frequency lines below 100 GHz) can be calculated from our centrifugal distortion analysis with high accuracy. This work also impacts work in both the fir and ir spectral

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regions. In the fir the results of our analysis of microwave data predict the higher- K_{-1} data of the fir to the accuracy of those data. These analyses also provide strong ground state constraints and starting points for several very interesting (2, 3) analyses of ir data now in progress.

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